Using Solubility Rules

- Soluble: will readily dissolve in water
- Insoluble: will not dissolve in water
- Slightly soluble: a tiny amount will dissolve in water

Soluble	Insoluble
Compounds	Compounds
NO₃⁻, Na⁺,	S ²⁻ , CO ₃ ²⁻ ,
K⁺, NH₄⁺	PO ₄ ³⁻
Cl ⁻ , Br, l ⁻	OH ⁻
(Ag ⁺ , Hg ²⁺ ,	(Na ⁺ , K ⁺ ,
Pb ²⁺)	Ca ²⁺)
SO ₄ ²⁻ (Ba ²⁺ , Pb ²⁺ , Ca ²⁺)	

Model 1 Rules Of Solubility In Aqueous Solutions Answers

Barron's Educational Series, Albert S. Tarendash

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Physical Chemistry for JEE Advanced: Part 2, 3E (Free Sample) K. S. Verma,2022-05-19 Physical Chemistry for JEE Advanced Part 2 a Cengage Exam Crack Series product is designed to help aspiring engineers focus on the subject of physical chemistry from two standpoints To develop their caliber aptitude and attitude for the engineering field and profession To strengthen their grasp and understanding of the concepts of the subjects of study and their applicability at the grassroots level Each book in this series approaches the subject in a very conceptual and coherent manner While its illustrative solved examples facilitate easy mastering of the concepts and their applications an array of solved problems exposes the students to a variety of questions that they can expect in the examination The coverage and features of this series of books make it highly useful for all those preparing for JEE Main and Advanced and aspiring to become engineers

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