

## SOLVED QUESTIONS

### Multiple Choice Questions

1.  $K_1$  and  $K_2$  are the velocity constants of forward and backward reactions. The equilibrium constant  $k$  of the reaction is
- (A)  $K_1 \times K_2$  (B)  $K_1 - K_2$   
 (C)  $\frac{K_1}{K_2}$  (D)  $\frac{K_1 + K_2}{K_1 - K_2}$
- Sol.** (C).
2. Which of the following reactions will be favoured at low pressure?
- (A)  $N_2 + O_2 \rightleftharpoons 2NO$  (B)  $H_2 + I_2 \rightleftharpoons 2HI$   
 (C)  $PCl_5 \rightleftharpoons 2PCl_3 + Cl_2$  (D)  $N_2 + 3H_2 \rightleftharpoons 2NH_3$
- Sol.** (C).
3. For the reaction,  $A + B \longrightarrow C + D$ , if active mass of A is kept constant and active mass of B is tripled, the rate of reaction will become
- (A) three times (B) six times  
 (C) eight times (D) nine times
- Sol.** (D).
4. The rate of  $\frac{K_p}{K_c}$  for the reaction:  
 $CO(g) + \frac{1}{2}O_2(g) \rightleftharpoons CO_2(g)$  is
- (A) 1 (B)  $RT$   
 (C)  $(RT)^{1/2}$  (D)  $(RT)^{3/2}$
- Sol.** (D).
5. Consider the reaction,  
 $CaCO_3(s) \rightleftharpoons CaO(s) + CO_2(g)$   
 in a closed container at equilibrium. What would be the effect of addition of  $CaCO_3$  on the equilibrium concentration of  $CO_2$ ?
- (A) Increases (B) Unpredictable  
 (C) Decreases (D) Remains unaffected
- Sol.** (D).
6. For the reaction,  $PCl_5(g) \rightleftharpoons PCl_3(g) + Cl_2(g)$ , the forward reaction at constant temperature is favoured by
- (A) introducing an inert gas at constant volume  
 (B) introducing  $PCl_3(g)$  at constant volume  
 (C) introducing  $PCl_5(g)$  at constant volume  
 (D) introducing  $Cl_2(g)$  at constant volume
- Sol.** (C).
7. A liquid is in equilibrium with its vapour at its boiling point. On an average, the molecules in the two phases have equal
- (A) intermolecular forces  
 (B) potential energy  
 (C) kinetic energy  
 (D) both kinetic energy and potential energy
- Sol.** (C).
8. The unit of equilibrium constant for the reversible reaction

# Multiple Choice Questions In Chemical Equilibrium

**J Ma**

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## **Multiple Choice Questions In Chemical Equilibrium:**

## The Enigmatic Realm of **Multiple Choice Questions In Chemical Equilibrium**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing in short supply of extraordinary. Within the captivating pages of **Multiple Choice Questions In Chemical Equilibrium** a literary masterpiece penned with a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of those that partake in its reading experience.

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### **Table of Contents Multiple Choice Questions In Chemical Equilibrium**

1. Understanding the eBook Multiple Choice Questions In Chemical Equilibrium
  - The Rise of Digital Reading Multiple Choice Questions In Chemical Equilibrium
  - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Questions In Chemical Equilibrium
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Multiple Choice Questions In Chemical Equilibrium
  - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Questions In Chemical Equilibrium
  - Personalized Recommendations
  - Multiple Choice Questions In Chemical Equilibrium User Reviews and Ratings
  - Multiple Choice Questions In Chemical Equilibrium and Bestseller Lists

5. Accessing Multiple Choice Questions In Chemical Equilibrium Free and Paid eBooks
  - Multiple Choice Questions In Chemical Equilibrium Public Domain eBooks
  - Multiple Choice Questions In Chemical Equilibrium eBook Subscription Services
  - Multiple Choice Questions In Chemical Equilibrium Budget-Friendly Options
6. Navigating Multiple Choice Questions In Chemical Equilibrium eBook Formats
  - ePub, PDF, MOBI, and More
  - Multiple Choice Questions In Chemical Equilibrium Compatibility with Devices
  - Multiple Choice Questions In Chemical Equilibrium Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Multiple Choice Questions In Chemical Equilibrium
  - Highlighting and Note-Taking Multiple Choice Questions In Chemical Equilibrium
  - Interactive Elements Multiple Choice Questions In Chemical Equilibrium
8. Staying Engaged with Multiple Choice Questions In Chemical Equilibrium
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Multiple Choice Questions In Chemical Equilibrium
9. Balancing eBooks and Physical Books Multiple Choice Questions In Chemical Equilibrium
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Multiple Choice Questions In Chemical Equilibrium
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Multiple Choice Questions In Chemical Equilibrium
  - Setting Reading Goals Multiple Choice Questions In Chemical Equilibrium
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Multiple Choice Questions In Chemical Equilibrium
  - Fact-Checking eBook Content of Multiple Choice Questions In Chemical Equilibrium
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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