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Operating Instructions for Dishwashers Miele, An instruction manual for the Miele dishwasher User manual Gesellschaft für Konsum-, Markt- und Absatzforschung, **Washing Machine WA 110, WA 115, WB 120, WB 125** Hoover,1997 First Edition Instruction Manual [washing Machine] Hotpoint, A user guide for the First Edition washing machine models WM51 WM52 WM53 DB/cockpit ,199? User's Manual UPD70108, UPD70116 NEC Electronics GmbH Düsseldorf,1985

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{cl aq} + 2\text{ag aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds Cl^- aq

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the complete ionic equation for this reaction is as follows equation 4 13 $2\text{Ag}^+ \text{aq} + 2\text{F}^- \text{aq} + 2\text{NH}_4^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow 2\text{Ag}_2\text{Cr}_2\text{O}_7 \text{s} + 2\text{NH}_4^+ \text{aq} + 2\text{F}^- \text{aq}$ because two NH_4^+ aq

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