

# AQA Chemistry

## GCSE Exam-style mark scheme

## C4 Chemical calculations

100 marks (20%)

1 hour

Question	Answer	Mark value (marks)	Mark	Mark available
10.1.1	1.4 g	1	1	100%
10.1.2	1.48 g	2	2	100%
		Answer: 1.48 g (1 mark) 1.4 g + 0.08 g (1 mark) <u>1.48 g</u> 100% as amount of 1.48 g. Examiner's marks should be indicated by 2 marks	2	100%
10.1.3	any 3 of: - weighing scales - wet litmus paper - colour - pH of the solution - any 3 of the above	3	3	100%
10.2	all 4 observations must be observed: - solution at 20°C is clear - white precipitate forms - white precipitate forms - white precipitate forms - white precipitate forms	4	4	100%
10.3.1	all 4 observations must be observed: - solution at 20°C is clear - white precipitate forms - white precipitate forms - white precipitate forms - white precipitate forms	4	4	100%
10.3.2	all 4 observations must be observed: - solution at 20°C is clear - white precipitate forms - white precipitate forms - white precipitate forms - white precipitate forms	4	4	100%
10.3.3	all 4 observations must be observed: - solution at 20°C is clear - white precipitate forms - white precipitate forms - white precipitate forms - white precipitate forms	4	4	100%

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**Jianjun Gao**



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