Part 2: Characteristics of Light

Try out your Spectrum Wheel by positioning the inner most of the flaps on EM SPECTRUM. This will simultaneously position the other flaps to ENERGY, WAVELENGTH & FREQUENCY.

Turn the upper wheel and observe the combinations.

Fill in the table below with the corresponding combinations you have observed using your Spectrum Wheel.

Table 1. Characteristics of Light

EM Spectnum	Energy	Frequency	Wavelength	Frequency x wavelength
Radio				
Microwave		100	1.1	
Infrared				
Visible Light				
Ultraviolet		100		
X-Ray		100	1 22	
Gamma Ray				

- Q1. How are frequency and wavelength related for a specific region of the spectrum?
- Q2. What can you observe with the values of the product of frequency and wavelength in the different spectra?
- Q3. How is ENERGY related to FREQUENCY?

Now that we are familiar with the electromagnetic spectrum and the corresponding energies, frequencies and wavelength probably we can see some applications of these in everyday living. UV rays are highly energetic than other spectral regions on its left. This could be a possible reason why we are not advised to stay under the sun after 9:00 in the morning. Prolong use of mobile phones may cause ear infection. This may be due to a higher energy emitted by microwaves used in cellular phones than radio waves commonly used in other communication devices. What about the visible spectrum? Do you want to know more about this spectral region?

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