

১৩.১. বিয়োজনের কণা পরমাণু পানি বিয়োজনে।

- (ক) ৫%
- (খ) ১৫%
- (গ) ৭০%
- (ঘ) ৯০%

১৩.২. কনসারভেটোর অসফল কণা বিচ্ছিন্নে পানিখর অর্ধমাত্রের দুগুণ কণা ছিল।

- (ক) 0.000004 cm (খ) 0.00004 cm
- (গ) 0.0004 mm (ঘ) 0.004 cm

১৩.৩. নিচের কোন মৌলটি অক্সিজেনে ইলেকট্রন গ্রহণের ক্ষমতা?

- (ক) Zn
- (খ) Fe
- (গ) Cu
- (ঘ) Pb

১৩.৪. পেরোয়টিকেশন অর্ধ কোষটির সবচেয়ে বেশি।

- (ক) Ba^{2+}
- (খ) Mg^{2+}
- (গ) Ca^{2+}
- (ঘ) Ba^{2+}

১৩.৫. AlF_3 এর প্রমাণ ০.০০০২ mol/L হলে প্রমাণ কণাকণা কণা।

- (ক) 3.4×10^{-14}
- (খ) 4.3×10^{-14}
- (গ) 3.4×10^{-13}
- (ঘ) 4.3×10^{-13}

১৩.৬. $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightleftharpoons 2\text{NH}_3(\text{g})$ বিক্রিতে K_p এর একক কোলটি।

- (ক) atm
- (খ) atm^2
- (গ) atm^{-1}
- (ঘ) atm^{-2}

	ক্রম \rightarrow	14	15
পরিমাণ \downarrow			
2		U	
3		V	W

[U, V এবং W বৌলের প্রতীক নয়।]

উল্লিখিত U, V ও W বৌলের ক্ষেত্রে—

- I. U এর ক্রোমাইড অর্ধ বিক্রি হইবে
- II. W পরিবর্তনশীল যোজনী দেখায়
- III. U ও W এর মৌল হইবে সাদৃশ্য বিয়োজন

নিচের কোনটি সঠিক?

- (ক) I ও II
- (খ) II ও III
- (গ) I ও III
- (ঘ) I, II ও III

উল্লিখিত হইবে ১৬ ও ২০০০ হইবে উল্লিখিত নয়।

	ক্রম \rightarrow	13	17
পরিমাণ \downarrow			
3		X	Y

[X ও Y বৌলের প্রতীক নয়।]

১৩.৭. X বৌলের অক্সাইডের অণুত্ব কণা।

- (ক) 2
- (খ) 4
- (গ) 6
- (ঘ) 8

১৩.৮. X ও Y বৌল কণা পানি বৌল—

- I. ক্রোমাইড পানি কণা
- II. কণা কণা কণা কণা কণা
- III. কণা কণা কণা কণা

নিচের কোনটি সঠিক?

- (ক) I ও II
- (খ) II ও III
- (গ) I ও III
- (ঘ) I, II ও III

১৩.৯. নিচের কোন পেরোয়টিকেশন সম-আয়ন প্রমাণ বিয়োজন?

- (ক) CaCl_2 , CH_3Cl
- (খ) H_2S , HCl
- (গ) C_2H_6 , HCl
- (ঘ) NaCl , $\text{C}_6\text{H}_5\text{Cl}$

১৩.১০. সবচেয়ে দুর্বল এসিড কোলটি?

- (ক) H_2SO_4
- (খ) HNO_3
- (গ) HClO_4
- (ঘ) H_3PO_3

১৩.১১. কলকোয়াল কণা কোলটি?

- (ক) 16
- (খ) 15
- (গ) 14
- (ঘ) 11

১৩.১২. সবচেয়ে কণার মূলনীতি কণাটি?

- (ক) 10
- (খ) 12
- (গ) 14
- (ঘ) 16

১৩.১৩. $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ বৌলে কণা কণার কণা বিয়োজন?

- (ক) 2
- (খ) 3
- (গ) 4
- (ঘ) 5

NM
CHE
MSE
TRY

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