

Roll No. _____

Code : 11201920PH-A

Please check that this question paper contains 37 questions and 8 printed pages.

Class : XI
PHYSICS (THEORY)

Time Allowed : 3 Hours

Annual Examination Feb.-March - 2020

Maximum Marks : 70

General Instructions :

1. All questions are compulsory. There are 37 questions in all.
2. This question paper has four sections : Section-A, Section-B, Section-C and Section-D.
3. Section-A contains twenty questions of one mark each.,
Section-B contains seven questions of two marks each.
Section-C contains seven questions of three marks each and
Section-D contains three questions of five marks each.
4. There is no overall choice. However, internal choices have been provided in two questions of one mark each, two questions of two marks each, one question of three marks and all the three questions of five marks weightage. You have to attempt only one of the choices in such questions.
5. Fifteen minutes time has been allotted to read this question paper. During this time, you will only read the question paper and will not write anything in the answer script.

SECTION - A

Direction (Q1-Q10) Select the most appropriate option from those given below each question.

1. The number of significant figures in 5400 is
(a) 3 (b) 4 (c) 2 (d) 1
2. The angle between $\vec{A} = (\hat{i} + \hat{j})$ and $\vec{B} = (\hat{i} - \hat{j})$ is
(a) 45° (b) 90° (c) -45° (d) 180°
3. A particle of mass 10 kg is moving in a straight line. The variation of its displacement, x with time t (in seconds) is given by $x = (t^3 - 2t - 10)\text{m}$. The force acting on it at $t = 4\text{s}$, is
(a) 24N (b) 240N (c) 300N (d) 480N
4. A body is falling freely under the action of gravity alone in vacuum. Which of the following quantities remain constant during the fall?
(a) kinetic energy (b) Potential energy
(c) Total mechanical energy (d) Total linear momentum

Format Of Physics Pquestion Paper For Grade 11

Victor M. Corman



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web log $[\text{H}^+]$ 10⁻⁴ 0.5 8.91 x 10⁻⁵ mol dm⁻³ log pK_a pH 3.75 4.05 0.30 10.0 3.0 0.50 and HCOOH 0.1 so HCOO^- 0.200 mol dm⁻³ amount H^+

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web only the better candidates could write successive ionisation equations in part c while part d about the pH of equimolar strong and weak acid solutions was poorly answered

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web to detect the presence of acids and bases micheal adds a few drops of phenolphthalein indicator to two colorless solutions a and b he observes the following changes in the

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web section 2 pH of strong acids number of protons released monoprotic acid acid that releases one H^+ ion per molecule e.g. HCl hydrochloric acid HNO_3 nitric acid CH_3COOH

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