

LANGUAGE PAPER 2 – SECTION B - COMPOSITION – 40 MARKS

Marking criteria for Section B

Table A, Composition: Content and structure

Level	Marks	General and specific marking criteria	
5	14–16	General	
		<ul style="list-style-type: none"> Content is complex, engaging and effective. (V1) Structure is secure, well balanced and carefully managed for deliberate effect. (V2) 	
		Specific – descriptive	Specific – narrative
		<i>Many well-defined and developed ideas and images create a convincing overall picture with varieties of focus.</i>	<i>The plot is well-defined and strongly developed with features of fiction writing such as description, characterisation and effective climax, and convincing details.</i>
4	11–13	General	
		<ul style="list-style-type: none"> Content is developed, engaging and effective. (V1) Structure is well managed, with some choices made for deliberate effect. (V2) 	
		Specific – descriptive	Specific – narrative
		<i>Frequent, well-chosen images and details give a mostly convincing picture.</i>	<i>The plot is defined and developed with features of fiction writing such as description, characterisation, climax and details.</i>
3	8–10	General	
		<ul style="list-style-type: none"> Content is relevant with some development. (V1) Structure is competently managed. (V2) 	
		Specific – descriptive	Specific – narrative
		<i>A selection of relevant ideas, images and details, even where there is a tendency to write in a narrative style.</i>	<i>The plot is relevant and cohesive, with some features such as characterisation and setting of scene.</i>
2	5–7	General	
		<ul style="list-style-type: none"> Content is straightforward and briefly developed. (V1) Structure is merely organised but may not always be effective. (V2) 	
		Specific – descriptive	Specific – narrative
		<i>The text is addressed with a series of relevant but straightforward details, which may be more typical of a narrative.</i>	<i>The plot is straightforward, with limited use of the features of narrative writing.</i>
1	3–4	General	
		<ul style="list-style-type: none"> Content is simple, and ideas and events may be limited. (V1) Structure is partially organised (but limited in its effect). (V2) 	
		Specific – descriptive	Specific – narrative
		<i>The recording of some relevant events with limited detail.</i>	<i>The plot is a simple narrative that may consist of events that are only partially linked and/or which are presented with partial clarity.</i>

Table B, Composition: Style and accuracy

Level	Marks	Description
5	21–24	<ul style="list-style-type: none"> Precise, well-chosen vocabulary and varied sentence structures, chosen for effect. (V3) Consistent well-chosen register suitable for the context. (V4) Spelling, punctuation and grammar almost always accurate. (V5)
4	17–20	<ul style="list-style-type: none"> Mostly precise vocabulary and a range of sentence structures mostly used for effect. (V3) Mostly consistent appropriate register suitable for the context. (V4) Spelling, punctuation and grammar mostly accurate, with occasional minor errors. (V5)
3	13–16	<ul style="list-style-type: none"> Some precise vocabulary and a range of sentence structures sometimes used for effect. (V3) Some appropriate register for the context. (V4) Spelling, punctuation and grammar generally accurate, but with some errors. (V5)
2	9–12	<ul style="list-style-type: none"> Simple vocabulary and a range of straightforward sentence structures. (V3) Simple register with a general awareness of the context. (V4) Frequent errors of spelling, punctuation and grammar, occasionally serious. (V5)
1	5–8	<ul style="list-style-type: none"> Limited and/or imprecise vocabulary and sentence structures. (V3) Limited and/or imprecise register for the context. (V4) Persistent errors of spelling, punctuation and grammar. (V5)
0	1–4	<ul style="list-style-type: none"> Frequently imprecise vocabulary and sentence structures. (V3) Register demonstrates little or no sense of the context. (V4) Persistent errors of spelling, punctuation and grammar impair communication. (V5)
0	0	<ul style="list-style-type: none"> No creditable content.

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web in today s experiment you will determine the solubility product ksp of calcium hydroxide Ca(OH)_2 by measuring the concentration of Ca(OH)_2 in a saturated solution calcium hydroxide is a sparingly soluble salt that dissolves according to the following reaction the solubility product expression for this reaction is

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web expert verified chem 182 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water a saturated aqueous solution of Ca(OH)_2 is represented in equation form as shown below $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$ the solubility product expression describes in mathematical terms the equilibrium

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web ksp for calcium hydroxide calcium hydroxide is a soft white caustic powder used in making mortar cements calcium salts paints and petrochemicals it is also used in saltwater aquaria to make up kalkwasser limewater solutions for reef tanks and is used as a ph regulating agent notice that calcium hydroxide is divalent and thus has twice

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web use the concentration of calcium hydroxide to calculate the concentration of calcium ions and hydroxide ions in the solution $\text{Ca(OH)}_2 \rightarrow \text{Ca}^{2+} + 2\text{OH}^{-}$ $5 \times 10^{-2} \text{ M}$ calcium ions $5 \times 10^{-2} \text{ M}$ $2 \times 10^{-2} \text{ M}$ hydroxide ions calculate calcium hydroxide ksp using the concentrations of calcium ions and hydroxide ions

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web madison bowe elana keller april 15 2014 introduction the objective of this lab was to measure the ksp of a slightly soluble basic salt and how solubility changes with temperature ksp is another form of equilibrium expression that deals with solubility the reaction that will take place is $\text{Ca(OH)}_2 \rightleftharpoons \text{Ca}^{2+} + 2\text{OH}^{-}$

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web question experiment 8 determining ksp for calcium hydroxide post lab report sheet data table 2 points trial 1 trial 2 hci molarity 0.050 M volume of filtered Ca(OH)_2 solution ml 5.00 5.00 3.71 13.75 equivalence point volume ml oh in saturated Ca(OH)_2 M Ca^{2+} in saturated Ca(OH)_2 M ksp of Ca(OH)_2 average ksp of Ca(OH)_2

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web the chemical equation used was $\text{Ca(OH)}_2(\text{s}) \rightleftharpoons \text{Ca}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq})$ the solubility product was determined to be $K_{sp} = [\text{Ca}^{2+}][\text{OH}^{-}]^2$ from the titration which was made by dripping a concentrated solution of calcium hydroxide into hydrochloric acid the calcium hydroxide was filtered before starting the titration

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web jul 12 2023 because the concentration of a pure solid such as $\text{Ca}_3(\text{PO}_4)_2$ is a constant it does not appear explicitly in the equilibrium constant expression the equilibrium constant expression for the dissolution of calcium phosphate is therefore

k Ca^{2+} 3PO_4^{3-} 4Ca^{2+} 2PO_4^{3-} 2Ca^{2+} 2PO_4^{3-} 2K^{+} Ca^{2+} 3PO_4^{3-} 4Ca^{2+}

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web lab report 8 determining of calcium hydroxide the purpose of this was to titrate a saturated $\text{Ca}(\text{OH})_2$ solution with a standard hcl solution and find the ksp or the OH^- concentration was found and then the concentration was found and u for this experiment a titration of calcium hydroxide with hydrochloric

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web question determining the ksp of calciumhydroxide $\text{Ca}(\text{OH})_2$ s Ca^{2+} aq 2OH^- aq for this experiment the titration of calciumhydroxide with hydrochloric acid was investigated determining the ksp of calcium hydroxide $\text{Ca}(\text{OH})_2$ s Ca^{2+} aq 2OH^- aq for this experiment the titration of calcium

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web question chem 182 experiment 8 determining the ksp of calcium hydroxide calcium hydroxide is an ionic solid that is sparingly soluble in water solution of $\text{Ca}(\text{OH})_2$ is represented in equation form lution of calcium ionic solid that is sparingly soluble in water a saturated aqueous as shown below

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web therefore it was determined through error calculations and ease of use that the most reliable valid method to use to get the closest ksp value to the theoretical ksp for calcium was the determination through ph method 2 introduction calcium hydroxide is a strong base soluble in water

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web plasma membrane homework symbiosis practice worksheet benzoin condensation grade a determining the ksp of calcium hydroxide after reading the lab complete items a b c and d title purpose chemicals and equipment and summary of procedure as described on page 10 of exp 1 on an 8 1/2 x 11 sheet of paper

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web question lab 9 determination of the solubility product ksp of calcium hydroxide introduction ionic compounds that are classified as insoluble based on solubility rules are actually slightly soluble each of these insoluble compounds actually dissolves to

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web we will filter the solid calcium hydroxide from the solution and then titrate the dissolved calcium hydroxide in solution with a known concentration of hydrochloric acid using titrimetric calculations we will determine the Ca^{2+} and OH^- concentrations in solution

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web calculate the K_{sp} for calcium hydroxide $K_{sp} Ca_2(OH)_2 = 2 \times 10^{-2}$ $K_{sp} 5 \times 10^{-4}$ find the accepted value of the K_{sp} for calcium hydroxide and compare it with your value discuss the discrepancy and suggest possible sources of experimental error
accepted value for K_{sp} for sodium hydroxide is 5×10^{-6}

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