

4. Attempt any THREE of the following:**12**

- a) Explain Dalton's law of partial pressure. How it is applicable to condenser?
- b) A system is composed of a gas contained in a cylinder fitted with a piston. The gas expands from the state 1 for which Internal energy $U_1 = 75$ KJ to state 2 for which $U_2 = -25$ KJ. During the expansion the gas does 60 KJ of work on the surrounding. Determine the heat transferred to or from the system during the process.
- c) 3 m^3 of gas of 30°C and 6 bar pressure is expanded isothermally to 1 bar. Find work done, change in internal energy and heat transferred during the process.
- d) Explain construction and working of shell and tube type heat exchanger. A ice plant producing 2000 Kg ice per day required the condenser. Suggest the type of condenser with justification.

5. Attempt any TWO of the following:**12**

- a)
 - (i) Suggest the methods to improve the performance of steam turbine. Explain any one in brief.
 - (ii) Identity the different losses occurred in steam turbine.
- b) An exterior wall of house consists 10.6 cm layer of common brick. It is followed by 3.8 cm layer of gypsum plaster and 5.83 cm of rockwool insulation. Estimate the amount of heat transferred through structure it.
Thermal conductivity of brick = 0.7 W/mk
Thermal conductivity of Plaster = 0.48 W/mk
Thermal conductivity of Insulation = 0.065 W/mk
- c) The initial condition of steam is 15% wet at a pressure of 7 bar. It expands to 1.2 bar by $PV^{1.3} = C$. Find
 - (i) Quality of steam at the end of expansion
 - (ii) Work done.

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