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web complete the changes in concentrations for each of the following reactions
$$\text{BaSO}_4 \text{ s} + \text{Ba}^{2+} \text{ aq} + \text{SO}_4^{2-} \text{ aq} \rightarrow \text{Ba}^{2+} \text{ aq} + \text{SO}_4^{2-} \text{ aq}$$
$$\text{Ag}_2\text{SO}_4 \text{ s} + 2 \text{Ag}^+ \text{ aq} + \text{SO}_4^{2-} \text{ aq} \rightarrow 2 \text{Ag}^+ \text{ aq} + \text{SO}_4^{2-} \text{ aq}$$

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