

# 1 • Chemical Foundations

1. A compound contains 40 g of carbon and 10 g of oxygen. It is classified as a hydrocarbon. Drawing these layers and filling in the rest of the structure, label and highlight what the compound is called. Name the structure and answer whether the compound is ionic or covalent. Write the formula of the compound for the given name.

- a. 40 g  
 b. 10 g  
 c. 50 g  
 d. 10 g

2. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.



3. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.
- a. CH4  
 b. CH3  
 c. CH2  
 d. CH
4. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.
- a. CH4  
 b. CH3  
 c. CH2  
 d. CH

5. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.



6. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.
- a. CH4  
 b. CH3  
 c. CH2  
 d. CH

7. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.

- a. CH4  
 b. CH3  
 c. CH2  
 d. CH

8. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.

- a. CH4  
 b. CH3  
 c. CH2  
 d. CH

9. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.

- a. CH4  
 b. CH3  
 c. CH2  
 d. CH

10. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.

- a. CH4  
 b. CH3  
 c. CH2  
 d. CH

11. A hydrocarbon is a compound that contains only carbon and hydrogen. Write the formula of the compound and name the compound.

- a. CH4  
 b. CH3  
 c. CH2  
 d. CH

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**F Rizvi**



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