

2007 Grade 11 Formula Sheet

Formulas that you may need to work questions on this test are found below.

You may refer to this page at any time during the mathematics test.

You may use calculator π or the number 3.14.

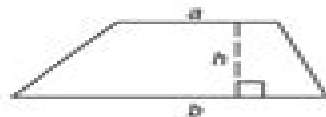


$$P = 2l + 2w$$

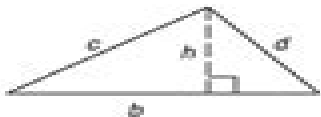
$$A = lw$$



$$A = bh$$



$$A = \frac{1}{2}h(a + b)$$



$$P = b + c + d$$

$$A = \frac{1}{2}bh$$



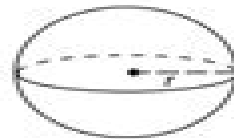
$$C = 2\pi r$$

$$A = \pi r^2$$



$$SA = 2lw + 2lh + 2wh$$

$$V = lwh$$



$$SA = 4\pi r^2$$

$$V = \frac{4}{3}\pi r^3$$



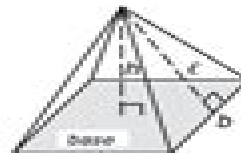
$$SA = 2\pi r^2 + 2\pi rh$$

$$V = \pi r^2 h$$



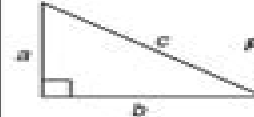
$$SA = \pi r^2 + \pi r \sqrt{r^2 + h^2}$$

$$V = \frac{1}{3}\pi r^2 h$$



$$SA = (\text{Area of the base}) + \frac{1}{2}l(\text{number of base sides})(b)$$

$$V = \frac{1}{3}(\text{Area of the base}) \times h$$



Pythagorean Theorem :

$$a^2 + b^2 = c^2$$

Permutations :

$$P(n, r) = \frac{n!}{(n-r)!}$$

Combinations :

$$C(n, r) = \frac{n!}{r!(n-r)!}$$

Distance Formula :

$$d = \sqrt{(x_2 - x_1)^2 + (y_2 - y_1)^2}$$

Midpoint :

$$\left(\frac{x_1 + x_2}{2}, \frac{y_1 + y_2}{2} \right)$$

Slope :

$$m = \frac{y_2 - y_1}{x_2 - x_1}$$

Point-Slope Formula :

$$(y - y_1) = m(x - x_1)$$

Slope-Intercept Formula :

$$y = mx + b$$

Standard Equation of a Line :

$$Ax + By = C$$

Mathematics Formula Chart 8th Grade

Reza Nazari, Ava Ross



Mathematics Formula Chart 8th Grade:

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web jul 19 2023 introduction titration is an analytical quantitative technique used to determine the concentration of a solute a ph titration is used to determine the concentration of an acid or a base titrations play an important role in determining amount and purity in many manufacturing processes

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{h}_2\text{c}_2\text{o}_4 + 2\text{naoh aq} \rightarrow \text{na}_2\text{c}_2\text{o}_4 + 2\text{h}_2\text{o}$ to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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web conclusion in this lab we used titration to explore the concepts of stoichiometry and equivalence points we found the concentration of an unknown substance by mixing 2m hcl with the naoh of unknown concentration in order to experimentally ascertain the concentration of the naoh

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction
 $\text{NaOH} + \text{C}_6\text{H}_4\text{COOH} \rightarrow \text{C}_6\text{H}_4\text{COONa} + \text{H}_2\text{O}$
l the point of equivalence is the point of titration when the number of moles of naoh na added is equal to the number of moles of khp nb in the