

ঘোর ও ঔষধীয় অণুসংকেত নির্দেশিত পরীক্ষার প্রয়োজক

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$W = 20 \text{ g}$$

$$HCl \text{ এর আপেক্ষিক ভর,}$$

$$M = 36.5 \text{ g/mol}$$

$$\text{চলকের আয়তন, } V = 1 \text{ L}$$

$$\text{চলকের ঘনমাত্রা } S = ?$$

উদাহরণে উল্লিখিত X-পদার্থটিতে 10 mg এর মধ্যে ১০০ পরিমাণ ঘন ও Y-পদার্থটিতে ০.১ g ১% $\left(\frac{W}{V}\right)$ HCl যখন নিচে কাজ করা হয়েছে।

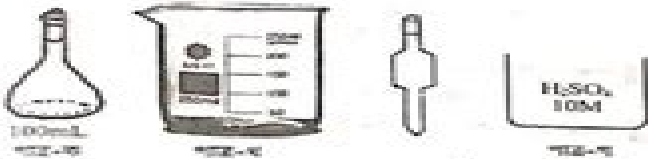
যখন X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও Y-পদার্থটি হলো ছোটো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিচে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও বিশুদ্ধতায় বিশুদ্ধতায় ব্যবহৃত হয়। এ পদ্ধতিতে খুব অল্প পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মৃদুপ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে বৈজ্ঞানিকভাবে ব্যবহার করা হয় ফলে পরিবেশে মৃদুপ ঘটে না।

$(CH_3NH_2 + H_2O \rightarrow CH_3NH_3^+ + OH^-)$ ফলে কোনো কোনো ক্ষেত্রে কম ঝুঁকির রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মৃদুপের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

উদাহরণ



১. উদাহরণে চলক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য যন্ত্রাঙ্ক প্রতিটিতে অঙ্কন করা এবং ব্যাখ্যা করা।
৪. উদাহরণের অণুসংকেত দ্রব্য থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রব্য তৈরি করার প্রতিস্থা ব্যাখ্যা করা। ৪ ও ৮ প্রশ্নের উত্তর

বিশ্লেষণের নিমিত্ত বীমের উপর রাইডার স্থাপন করলে বীমের প্রতি স্কেলের জন্য যে ছবি পাওয়া যায়, তাকে রাইডার ছবি বলা হয়।

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও বিশুদ্ধতায় বিশুদ্ধতায় ব্যবহৃত হয়। এ পদ্ধতিতে খুব অল্প পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মৃদুপ উল্লেখযোগ্যভাবে কমে আসে।

অতএব, কোনো কোনো ক্ষেত্রে কম ঝুঁকির রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মৃদুপের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

প্রতিক্রিয়া	ফলস্বরূপ (Corrosive)
আয়তন	ফলস্বরূপে রাসায়নিক উপাদান কমে যে ফলস্বরূপে রাসায়নিক দ্রব্যকে খুব অল্প পরিমাণে ব্যবহার করা হয়। এতে খুব খুঁচু দ্রব্য এবং অন্যদিকে পদার্থ পড়ে যায়।
উপাদান	খুঁচু দ্রব্যের এমিউলসিওন (HCl, HNO ₃ , H ₂ SO ₄) পাত্র ফলে (NaOH, KOH, H ₂ O ₂) আয়তনমিত লিকার প্রভৃতি মৃদুপ ইত্যাদি।
সংরক্ষণ	- খুঁচু এমিউলসিওন পাত্রের মধ্যে রাখা হয়। - খুঁচু ফলস্বরূপে, ফলস্বরূপে পাত্রের পাত্রের মধ্যে রাখা হয়।

উদাহরণের মূল্যটি হলো 10M H_2SO_4 দ্রব্য এবং এর থেকে 1M ঘনমাত্রার দ্রব্য প্রস্তুত করতে হবে 1ম ও ১০০ মিলি লিটারের মধ্যে।

1ম লিটারের মধ্যে 100 mL আয়তনমিত দ্রব্য ও এর 100 mL লিটারে। যেহেতু আয়তনমিত দ্রব্যের আয়তন 100 mL, তাই আমাদের 100 mL 1M ঘনমাত্রার H_2SO_4 দ্রব্য তৈরি করতে হবে।

$$V_1 S_1 = V_2 S_2$$

$$\text{যা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10mL H_2SO_4 দ্রব্যের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL।

আবার, লিটারের মধ্যে 10 mL দ্রব্যের ঘনমাত্রার দ্রব্য নিচে তার সাথে আয়তনমিত দ্রব্য 90 mL পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার দ্রব্য পাওয়া যায়।



১. অণুসংকেতটি কাকে বলা হয়?
২. সাদা ছবি K₂ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা করা।
৩. A ডিফের হেটিকের উত্তর করার কৌশল বর্ণনা করা।
৪. B ও C ডিফের সাথে সম্পর্কিত রাসায়নিক দ্রব্যের ব্যবহারকালে খাখা ঝুঁকি হলে কবীর উপর সম্পর্কিত বিশ্লেষণ করা। ৪ ও ৮ প্রশ্নের উত্তর

অণুসংকেতটি এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সংকেত যা মৃদুপ বা উপাদানসমূহকে জারণ প্রতিস্থাপন কমে থেকে রাখা করে রাখার সংকেত।

সামান্য K₂ এর ঘন কখনো অসীম বা শূন্য হতে পারে না। তাইটি ব্যাখ্যা করার জন্য নিচের উত্তরদুই মাইক্রোপরি বিশ্লেষণ করি-

$$A + B \rightarrow C + D$$

Hsc Chemistry 1 First Chapter Notes

Sabine Zange



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