

How Substances Dissolve

KEY IDEAS

As you read this section, keep these questions in mind:

- Why is water called the universal solvent?
- How do substances dissolve?

Why Do Substances Dissolve in Certain Solvents?

Water is often called a universal solvent because many substances dissolve in it. However, no one substance can dissolve every solute. A general rule in chemistry is that “like dissolves like.” This rule means that a solvent will dissolve substances that have similar molecular structures. 

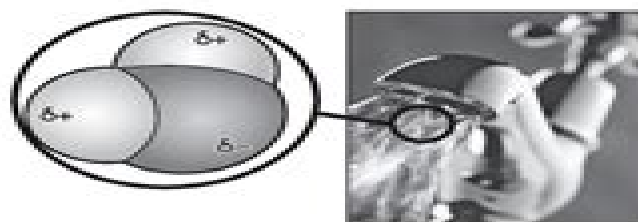
A solvent must be able to attract solute particles and pull them away from one another. Thus, solvent particles must attract solute particles more strongly than the solute particles attract one another.

POLAR MOLECULES

Recall that atoms bonded together do not all share electrons equally. For example, a water molecule is made up of two hydrogen atoms bonded to one oxygen atom. The oxygen atom attracts electrons more strongly than the hydrogen atoms do. Thus, the electrons spend more time near the oxygen atom. This gives each hydrogen atom a partial positive charge (δ^+) and the oxygen atom a partial negative charge (δ^-). 

Molecules that have partially charged positive and negative areas are **polar**. Because water molecules are polar, water can dissolve many other polar substances. Water can also dissolve ionic compounds such as table salt.

Water is a polar molecule because the oxygen atom strongly attracts electrons. Because the electrons spend more time near the oxygen atom, the oxygen atom has a partial negative charge.



READING TOOLBOX

Summarize Read each page of the section silently to yourself. With a partner, take turns summarizing each page.

READING CHECK

1. Identify Why is water called a universal solvent?

because
many substances dissolve
in it

READING CHECK

2. Explain Why does a water molecule have partial charges?

The oxygen atom attracts
electrons more strongly
than the hydrogen atoms
do. Thus, the electrons
spend more time near the
oxygen atom.

How Substances Dissolve Answer Key

RJ Shavelson



How Substances Dissolve Answer Key:

Whispering the Techniques of Language: An Mental Quest through **How Substances Dissolve Answer Key**

In a digitally-driven world wherever monitors reign great and quick conversation drowns out the subtleties of language, the profound secrets and mental subtleties concealed within phrases usually go unheard. Yet, nestled within the pages of **How Substances Dissolve Answer Key** a captivating literary treasure pulsing with fresh emotions, lies a fantastic journey waiting to be undertaken. Published by an experienced wordsmith, that charming opus invites viewers on an introspective trip, lightly unraveling the veiled truths and profound influence resonating within the fabric of every word. Within the mental depths of the moving review, we will embark upon a sincere exploration of the book is core subjects, dissect its captivating publishing type, and yield to the effective resonance it evokes heavy within the recesses of readers hearts.

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