

SCQ (Single Correct Type) :

- Nuclei of radioactive element A are produced at rate ' P ' at any time t . The element A has decay constant λ . Let N be the number of nuclei of element A at any time t . At time $t = t_0$, $\frac{dN}{dt}$ is minimum. Then the number of nuclei of element A at time $t = t_0$ is

(A) $\frac{\lambda t_0^2 - 2t_0}{\lambda^2}$ (B) $\frac{t_0 - \lambda t_0^2}{\lambda^2}$ (C) $\frac{2t_0 - \lambda t_0^2}{\lambda}$ (D) None of these
- A radioactive sample decays by β -emission. In first two seconds ' n ' β -particles are emitted and in next 2 seconds, ' $0.25n$ ' β -particles are emitted. The half life of radioactive nuclei is

(A) 2 sec (B) 4 sec (C) 1 sec (D) none
- A factory produces a radionuclide at a constant rate R . Initially there were no nuclide present. Then activity of the sample produced by factory just after an average life of nuclide passes away is :

(A) R (B) Re^{-1} (C) $R(1 - e^{-1})$ (D) $\frac{R}{2} (1 - e^{-1})$
- A radionuclide with decay constant λ is produced in a reactor at a constant rate α nuclei per second. During each decay energy E_0 MeV is released. If the production of radio nuclides started at $t = 0$, total energy released upto time t is given by equation

(A) $E_0 \alpha (1 - e^{-\lambda t})$ (B) $E_0 \frac{\alpha}{\lambda} (1 - e^{-\lambda t})$ (C) $E_0 (\alpha t - \frac{\alpha}{\lambda} e^{-\lambda t})$ (D) $E_0 \left(\alpha t - \frac{\alpha}{\lambda} (1 - e^{-\lambda t}) \right)$
- The count rate observed from a radioactive source at ' t ' second was N_0 and at $4t$ second it was $\frac{N_0}{16}$. The count rate observed, at $\left(\frac{11}{2}\right)t$ second will be

(A) $\frac{N_0}{128}$ (B) $\frac{N_0}{64}$ (C) $\frac{N_0}{32}$ (D) None of these
- The count rate from 100 cm^3 of a radio-active liquid is C . Some of this liquid is now discarded. The count rate of the remaining liquid is found to be $C/10$ after three half-lives. The volume of the remaining liquid in cm^3 is

(A) 20 (B) 40 (C) 60 (D) 80

Grade 1exam Papers Physics Caps Exemplar

R Sandford



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