

AQA Chemistry

GCSE Exam-style mark scheme

C4 Chemical calculations

100 marks (20%)

1 hour

Question	Answer	Mark	Mark	Mark
10.1.1	14.2 g		1	100
10.1.2	14.2 g	<p>Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$</p>	10	100
10.1.3	<p>any 1000 factor</p> <ul style="list-style-type: none"> converting grams into kg (multiplication) needed into kg (multiplication) needed into the equation 		10	100
10.2	<p>all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct</p>	<p>Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$</p>	10	100
10.3.1	<p>all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct</p>	<p>Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$</p>	10	100
10.3.2	<p>all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct</p>	<p>Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$</p>	10	100
10.3.3	<p>all calculations must be shown amounts all used in the calculation all units must be correct must be correct must be correct must be correct must be correct</p>	<p>Amount of magnesium carbonate $14.2 \text{ g} \div 142.2 \text{ g mol}^{-1}$ $= 0.1 \text{ mol}$ $\times 142.2 \text{ g mol}^{-1}$ $= 14.2 \text{ g}$</p>	10	100

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