

Assignment #2 – Compound Names and Formulas

Multi-valent ions only

A. Write the correct formula for the following compounds, all of which have been named using the modern Roman Numeral Method. The ionic charge is given after the first element (metallic) in Roman Numerals.

1. copper (II) oxide CuO
2. mercury (I) oxide Hg₂O
3. gold (III) chloride AuCl₃
4. thallium (III) bromide TlBr₃
5. bismuth (V) oxide Bi₂O₅
6. terbium (III) oxide Tb₂O₃
7. uranium (VI) oxide UO₃
8. protactinium (IV) bromide PaBr₄
9. cerium (III) oxide Ce₂O₃
10. arsenic (V) sulphide As₂S₅

11. manganese (III) oxide Mn₂O₃
12. vanadium (II) bromide VBr₂
13. niobium(V) oxide Nb₂O₅
14. titanium (III) oxide Ti₂O₃
15. titanium (III) nitride TiN
16. iron (II) oxide FeO
17. cobalt (II) phosphide Co₃P₂
18. tin (II) oxide SnO
19. thulium (II) bromide TmBr₂
20. copper (II) bromide CuBr₂

B. Name the compound using the Roman Numeral Method and balance the ionic charges.

1. SnCl₄ Tin (IV) Chloride
2. BiBr₅ Bismuth (V) Bromide
3. PoO₂ Polonium (IV) Oxide
4. PbI₂ Lead (II) Iodide
5. HgO Mercury (II) Oxide
6. HgCl Mercury (I) Chloride
7. Au₂O₃ Gold (III) Oxide
8. FeCl₂ Iron (II) Chloride

9. PdF₄ Palladium (IV) Fluoride
10. Os₂O₃ Osmium (III) Oxide
11. MoBr₂ Molybdenum (II) Bromide
12. VCl₅ Vanadium (V) Chloride
13. Mn₂O₃ Manganese (III) Oxide
14. CoO Cobalt (II) Oxide
15. Np₂O₃ Neptunium (III) Oxide
16. V₂O₅ Vanadium (V) Oxide

Ionic Compounds Homework Review Sheet Answer Key

Daniel F McAuley



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Table of Contents Ionic Compounds Homework Review Sheet Answer Key

1. Understanding the eBook Ionic Compounds Homework Review Sheet Answer Key
 - The Rise of Digital Reading Ionic Compounds Homework Review Sheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Compounds Homework Review Sheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Compounds Homework Review Sheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Compounds Homework Review Sheet Answer Key
 - Personalized Recommendations
 - Ionic Compounds Homework Review Sheet Answer Key User Reviews and Ratings
 - Ionic Compounds Homework Review Sheet Answer Key and Bestseller Lists
5. Accessing Ionic Compounds Homework Review Sheet Answer Key Free and Paid eBooks
 - Ionic Compounds Homework Review Sheet Answer Key Public Domain eBooks
 - Ionic Compounds Homework Review Sheet Answer Key eBook Subscription Services
 - Ionic Compounds Homework Review Sheet Answer Key Budget-Friendly Options

6. Navigating Ionic Compounds Homework Review Sheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Compounds Homework Review Sheet Answer Key Compatibility with Devices
 - Ionic Compounds Homework Review Sheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Compounds Homework Review Sheet Answer Key
 - Highlighting and Note-Taking Ionic Compounds Homework Review Sheet Answer Key
 - Interactive Elements Ionic Compounds Homework Review Sheet Answer Key
8. Staying Engaged with Ionic Compounds Homework Review Sheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Compounds Homework Review Sheet Answer Key
9. Balancing eBooks and Physical Books Ionic Compounds Homework Review Sheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Compounds Homework Review Sheet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Ionic Compounds Homework Review Sheet Answer Key
 - Setting Reading Goals Ionic Compounds Homework Review Sheet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ionic Compounds Homework Review Sheet Answer Key
 - Fact-Checking eBook Content of Ionic Compounds Homework Review Sheet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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