

## NOMENCLATURE PACKET

### Worksheet I: Binary Ionic Compounds (representative metals)

1. Name the following ionic compounds:

- a.  $\text{Al}_2\text{O}_3$  Aluminum oxide
- b.  $\text{Cs}_2\text{O}$  Cesium oxide
- c.  $\text{Rb}_3\text{N}$  Rubidium nitride
- d.  $\text{Ca}_3\text{N}_2$  Calcium nitride
- e.  $\text{SrSe}$  Strontium selenide
- f.  $\text{Cs}_2\text{S}$  Cesium sulfide
- g.  $\text{Al}_2\text{S}_3$  Aluminum sulfide
- h.  $\text{LiBr}$  Lithium bromide
- i.  $\text{Mg}_3\text{N}_2$  Magnesium nitride
- j.  $\text{CaF}_2$  Calcium fluoride

2. Write the chemical formula for the following ionic compounds:

- a. Barium nitride  $\text{Ba}_3\text{N}_2$
- b. Indium Fluoride  $\text{InF}_3$
- c. Calcium oxide  $\text{CaO}$
- d. Sodium nitride  $\text{Na}_3\text{N}$
- e. Magnesium chloride  $\text{MgCl}_2$
- f. Potassium oxide  $\text{K}_2\text{O}$
- g. Magnesium oxide  $\text{MgO}$
- h. Potassium sulfide  $\text{K}_2\text{S}$
- i. Lithium nitride  $\text{Li}_3\text{N}$
- j. Strontium fluoride  $\text{SrF}_2$
- k. Aluminum sulfide  $\text{Al}_2\text{S}_3$
- l. Calcium oxide  $\text{CaO}$

# Ionic Review Answer Key

**Liyong Dong**

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