## ChemThink: Covalent Bonding

Covalent bonding forms when atoms are				electrons.
	atoms get close end / Neutrons / Elect		is attracts th	e other atom's
	ding, the atom's ele			
electrons spend most of their time				
	t be able to hold on anoth			, while
Covalent bo	onds form between	two		
At a certain and move t	s move closer, the point the potential he atoms closer be	energy		if you try
The ideal d	istance between the	e atoms is know		·
Bond Type	Draw an Example	# of paired e	Total # of e shared	Strongest/ Weakest
Single				
Double				
Triple				

## Naming Simple Covalent Compounds:

The ending of the name of the second element is changed to \_\_\_\_\_\_.

## **Ionic Bonds Form When Chemthink**

**Tao Wei** 

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