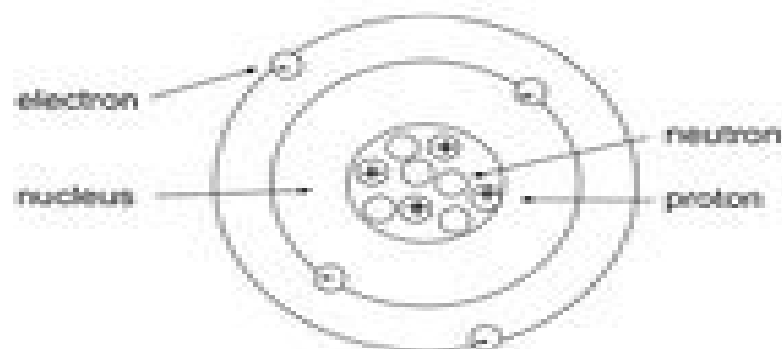


WHAT IS (the) MATTER?

Matter is anything that has mass and takes up space (volume). Matter is made up of small particles, known as atoms.

Diagram of an Atom



Atomic mass = number of _____ +
number of _____

Atomic number = number of _____

In a neutral atom,
the number of **electrons** (charge -1) is the
same as the number of **protons** (charge +1)

Problem 1: Chromium has an atomic mass of 52 and an atomic number of 24. Fill in the following:

Number of protons:

Number of neutrons:

Number of electrons:

Problem 2: Tin has an atomic mass of 118 and an atomic number of 50. Fill in the following:

Number of protons:

Number of neutrons:

Number of electrons:

Part 2: Introduction to Ions

An ion is a charged atom. How does an atom acquire a charge? When the number of electrons do not equal the number of protons, this creates a charge. Why? Remember,

Ion Practice Skill Practice 18 Answers

LL Leslie



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Reviewing **Ion Practice Skill Practice 18 Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Ion Practice Skill Practice 18 Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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