

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.
 2. Write the symbol for the ion that has the same number of protons as the element but a different number of electrons.

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained	Charge
1	Sodium	Na^+	11	10	lost one	positive
2	Chlorine	Cl^-	17	18	gained one	negative
3	Strontium	Sr^{2+}	38	36	lost two	positive
4	Polonium	Po^{2-}	84	86	gained two	negative
5	Calcium	Ca^{2+}	20	18	lost two	positive
6	Barium	Ba^{2+}	56	54	lost two	positive
7	Hydrogen	H^+	1	0	lost one	positive
8	Lithium	Li^+	3	2	lost one	positive
9	Fluorine	F^-	9	10	gained one	negative
10	Oxygen	O^{2-}	8	10	gained two	negative
11	Sulfur	S^{2-}	16	18	gained two	negative
12	Aluminum	Al^{3+}	13	10	lost three	positive
13	Bromine	Br^-	35	36	gained one	negative
14	Iron	Fe^{2+}	26	24	lost two	positive
15	Gold	Au^+	79	78	lost one	positive
16	Strontium	Sr^{2+}	38	36	lost two	positive
17	Calcium	Ca^{2+}	20	18	lost two	positive

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