

Chemistry Notes - Grade 10

5.1 Properties and Changes

Physical Change: Water to Ice

Chemical Change: Hydrogen peroxide to Water

Physical Property:

- ↳ Characteristic/description of the substance
- ↳ Based on Qualitative Observations (senses)
- ↳ Also based on Quantitative Observations (calculations and measurements) (e.g. colour, texture, density, smell)

Chemical Property:

- ↳ Describes how a substance behaves in the presence of another substance (e.g. how it reacts with other elements and compounds)
- ↳ Examples are flammability, corrosion, bleaching, rusting etc.

Physical Change:

- ↳ No new substance created
- ↳ Most can be reversed
- ↳ E.g. Change of state, dissolving sugar in water, cutting paper

Chemical Change:

- ↳ Produces a new substance
- ↳ Most *can't* be reversed
- ↳ E.g. Forest fire, boiling an egg, burning wood

Five Clues that a Chemical Change has Occurred:

1. A new colour is observed
2. A new odour is produced
3. Heat or light is produced or absorbed
4. Bubbles of gas are formed
5. A solid material (precipitate) forms in a liquid

5.4 Patterns in the Periodic Table

Where are the metals, non metals, and metalloids located on the periodic table?

- ↳ *Metals:* Left side
- ↳ *Metalloid:* On the staircase
- ↳ *Non Metals:* Right side exception being Hydrogen

Where are the alkali metals found? What are their properties?

- ↳ *Location:* Group 1
- ↳ *Properties:* Highly Reactive, shiny, soft, low melting point, and they have one valence electrons, soft, ductile

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Annelies Wilder-Smith



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Table of Contents Grade Chemistry Notes

1. Understanding the eBook Grade Chemistry Notes
 - The Rise of Digital Reading Grade Chemistry Notes
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade Chemistry Notes
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade Chemistry Notes
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade Chemistry Notes
 - Personalized Recommendations
 - Grade Chemistry Notes User Reviews and Ratings
 - Grade Chemistry Notes and Bestseller Lists

5. Accessing Grade Chemistry Notes Free and Paid eBooks
 - Grade Chemistry Notes Public Domain eBooks
 - Grade Chemistry Notes eBook Subscription Services
 - Grade Chemistry Notes Budget-Friendly Options
6. Navigating Grade Chemistry Notes eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade Chemistry Notes Compatibility with Devices
 - Grade Chemistry Notes Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade Chemistry Notes
 - Highlighting and Note-Taking Grade Chemistry Notes
 - Interactive Elements Grade Chemistry Notes
8. Staying Engaged with Grade Chemistry Notes
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade Chemistry Notes
9. Balancing eBooks and Physical Books Grade Chemistry Notes
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade Chemistry Notes
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Grade Chemistry Notes
 - Setting Reading Goals Grade Chemistry Notes
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Grade Chemistry Notes
 - Fact-Checking eBook Content of Grade Chemistry Notes
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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