

PART - I

15 X 1 = 15

I Answer all the questions.

1. Which of the following is / are true with respect to Carbon - 12
a) relative atomic mass is 12u b) oxidation number of Carbon is +4 in all its compounds
c) mole of Carbon - 12 contain 6.022×10^{22} carbon atoms d) all of these
2. How many electrons in an atom with atomic number 105 can have $(n + l) = 8$
a) 30 b) 17 c) 15 d) unpredictable
3. Among the following which is isoelectronic with O^{2-} ion,
a) Cl^- b) F^- c) Ca^{2+} d) N^{2-}
4. The cause of permanent hardness of water is due to
a) $Ca(HCO_3)_2$ b) $Mg(HCO_3)_2$ c) $CaCl_2$ d) $MgCO_3$
5. Sodium is stored in
a) Alcohol b) Water c) Kerosene d) None of these.
6. If temperature and volume of an ideal gas is increased to twice its values, the initial pressure P becomes
a) 4p b) 2p c) p d) 3p
7. Heat of combustion is always
a) Positive b) negative c) zero d) either positive or negative
8. In the reaction, $Fe(OH)_3(s) \rightleftharpoons Fe^{3+}(aq) + 3OH^-(aq)$ if the concentration of OH^- ions is decreased by $1/4$ times, then the equilibrium concentration of Fe^{3+} will
a) not changed b) also decreased by $1/4$ times
c) increase by 4 times d) increase by 64 times
9. According to Raoult's law, the relative lowering of vapour pressure for a solution is equal to
a) mole fraction of solvent b) mole fraction of solute
c) number of moles of solute d) number of moles of solvent
10. Number of sigma (σ) and pi (π) bonds in acetylene is
a) 2σ & 2π b) 3σ & 3π c) 2σ & 3π d) 3σ & 2π
11. Sodium nitropruside reacts with sulphide ion to give a purple colour due to the formation of
a) $[Fe(CN)_5 NO]^{3-}$ b) $[Fe(NO)_5 CN]^+$ c) $[Fe(CN)_5 NOS]^{4-}$ d) $[Fe(CN)_5 NOS]^{3-}$
12. What is the hybridisation state of benzyl carbonium ion?
a) sp^2 b) sp^3 c) sp^2 d) sp^2
13. is used for welding and cutting metals.
a) Ethylene b) Propylene c) Poly ethylene d) Oxy - acetylene
14. What should be the correct IUPAC name of diethyl chloromethane?
a) 3 - chloro pentane b) 1 - chloropentane
c) 1 - chloro - 1, 1, diethyl methane d) 1 - chloro - 1 - ethylepropane
15. The pH of normal rain water is
a) 6.5 b) 7.5 c) 5.6 d) 4.6

PART - II

6 x 2 = 12

II Answer any six. (q.No. 24 is compulsory).

16. Define modern periodic law.
17. What is water - gas shift reaction?

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Table of Contents Grade 1chemistry Half Yearly Exam

1. Understanding the eBook Grade 1chemistry Half Yearly Exam
 - The Rise of Digital Reading Grade 1chemistry Half Yearly Exam
 - Advantages of eBooks Over Traditional Books
2. Identifying Grade 1chemistry Half Yearly Exam
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Grade 1chemistry Half Yearly Exam
 - User-Friendly Interface
4. Exploring eBook Recommendations from Grade 1chemistry Half Yearly Exam

-
- Personalized Recommendations
 - Grade 1chemistry Half Yearly Exam User Reviews and Ratings
 - Grade 1chemistry Half Yearly Exam and Bestseller Lists
5. Accessing Grade 1chemistry Half Yearly Exam Free and Paid eBooks
 - Grade 1chemistry Half Yearly Exam Public Domain eBooks
 - Grade 1chemistry Half Yearly Exam eBook Subscription Services
 - Grade 1chemistry Half Yearly Exam Budget-Friendly Options
 6. Navigating Grade 1chemistry Half Yearly Exam eBook Formats
 - ePub, PDF, MOBI, and More
 - Grade 1chemistry Half Yearly Exam Compatibility with Devices
 - Grade 1chemistry Half Yearly Exam Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Grade 1chemistry Half Yearly Exam
 - Highlighting and Note-Taking Grade 1chemistry Half Yearly Exam
 - Interactive Elements Grade 1chemistry Half Yearly Exam
 8. Staying Engaged with Grade 1chemistry Half Yearly Exam
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Grade 1chemistry Half Yearly Exam
 9. Balancing eBooks and Physical Books Grade 1chemistry Half Yearly Exam
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Grade 1chemistry Half Yearly Exam
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Grade 1chemistry Half Yearly Exam
 - Setting Reading Goals Grade 1chemistry Half Yearly Exam
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Grade 1chemistry Half Yearly Exam

- Fact-Checking eBook Content of Grade 1chemistry Half Yearly Exam
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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