

**MARKING SCHEME**  
**SAMPLE PAPER (2024 -25)**  
**CHEMISTRY THEORY**  
**(043)**

<b>SECTION A</b>		
1	(c) reacts with Benzenesulphonyl chloride to form a product that is insoluble in alkali	1
2	(b) $\text{CH}_3\text{Cl}$ The order followed is this $\text{CH}_3\text{I} < \text{CH}_3\text{Br} < \text{CH}_3\text{F} < \text{CH}_3\text{Cl}$ , though F is most electronegative, the bond length is small as compared to C-Cl. Dipole moment is the product of the charge and the bond length.	1
3	(a) (i)-(C), (ii)-(B), (iii)-(A)	1
4	(d) 1 =Bromomethane, 2= 2-Bromo-2-methylpropane, 3=2-Bromobutane, 4= 1-Bromobutane  <b>(for visually challenged learners)</b> d. 1-Bromobutane	1  1
5	(c) the order of reaction is zero as the unit of k is $\text{molL}^{-1}\text{s}^{-1}$ . Thus half life = $[\text{R}]_0 / 2k = 4.62 \times 10^{-2} / 2 \times 2.31 \times 10^{-2}$	1
6	(b) Benzoic acid and ethanoic acid  $\text{C}_6\text{H}_5\text{COOCOCH}_3 \xrightarrow{\text{H}_2\text{O}} \text{C}_6\text{H}_5\text{COOH} + \text{CH}_3\text{COOH}$	1
7	(b) $\text{X}=[\text{Co}(\text{NH}_3)_4\text{Cl}_2]^+\text{Cl}^-$ , Y= 1:3	1
8	(b) Cellulose Starch contains only $\alpha$ glucose, sucrose contains $\alpha$ -D-glucose and $\beta$ -D-fructose glucose, maltose contains $\alpha$ -D-glucose and cellulose is a polymer of $\beta$ -D-glucose.	1
9	(d) $\text{Ti}^{3+} < \text{Cr}^{3+} < \text{Fe}^{2+} < \text{Mn}^{2+}$ No. of unpaired electrons : $\text{Ti}^{3+}$ (1), $\text{Cr}^{3+}$ (3), $\text{Fe}^{2+}$ (4) and $\text{Mn}^{2+}$ (5) Paramagnetism depends on the number of unpaired electrons	1
10	(d) It never goes to completion First order reaction $[\text{R}] = [\text{R}_0] e^{-kt}$ If $[\text{R}]=0$ then	1

# K94749 Chemitry Mark Scheme

**Camilla Rothe**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

**K94749 Chemitry Mark Sheme:**

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