

- Directions:**
 1. An element may be assigned a positive or negative charge depending on whether it is a metal or nonmetal.
 2. An atom with a positive charge is a cation.
 3. An atom with a negative charge is an anion.
 4. The number of protons in an atom is the atomic number.

Ion Practice Set

1. Write the symbol for the ion that has the following number of protons and electrons:
 11 protons, 10 electrons
 17 protons, 18 electrons
 19 protons, 18 electrons
 20 protons, 18 electrons
 21 protons, 18 electrons
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 90 protons, 18 electrons
 91 protons, 18 electrons
 92 protons, 18 electrons
 93 protons, 18 electrons
 94 protons, 18 electrons
 95 protons, 18 electrons
 96 protons, 18 electrons
 97 protons, 18 electrons
 98 protons, 18 electrons
 99 protons, 18 electrons
 100 protons, 18 electrons

Complete the following table, using the periodic table in the back of your book.

	Element Name	Ion Symbol	Number of Protons	Number of Electrons	Number of Electrons Lost or Gained	Charge
1	Sodium	Na^+	11	10	lost one	positive
2	Chlorine	Cl^-	17	18	gained one	negative
3	Calcium	Ca^{2+}	20	18	lost two	positive
4	Oxygen	O^{2-}	8	10	gained two	negative
5	Aluminum	Al^{3+}	13	10	lost three	positive
6	Sulfur	S^{2-}	16	18	gained two	negative
7	Hydrogen	H^+	1	0	lost one	positive
8	Lithium	Li^+	3	2	lost one	positive
9	Fluorine	F^-	9	10	gained one	negative
10	Strontium	Sr^{2+}	38	36	lost two	positive
11	Barium	Ba^{2+}	56	54	lost two	positive
12	Hydrogen	H^+	1	0	lost one	positive
13	Lithium	Li^+	3	2	lost one	positive
14	Fluorine	F^-	9	10	gained one	negative
15	Sulfur	S^{2-}	16	18	gained two	negative
16	Calcium	Ca^{2+}	20	18	lost two	positive
17	Oxygen	O^{2-}	8	10	gained two	negative
18	Aluminum	Al^{3+}	13	10	lost three	positive
19	Sulfur	S^{2-}	16	18	gained two	negative
20	Calcium	Ca^{2+}	20	18	lost two	positive
21	Scandium	Sc^{3+}	21	18	lost three	positive
22	Titanium	Ti^{4+}	22	18	lost four	positive
23	Vanadium	V^{5+}	23	18	lost five	positive
24	Chromium	Cr^{3+}	24	21	lost three	positive
25	Manganese	Mn^{2+}	25	23	lost two	positive
26	Iron	Fe^{2+}	26	24	lost two	positive
27	Cobalt	Co^{3+}	27	24	lost three	positive
28	Nickel	Ni^{2+}	28	26	lost two	positive
29	Copper	Cu^+	29	28	lost one	positive
30	Zinc	Zn^{2+}	30	28	lost two	positive
31	Gallium	Ga^{3+}	31	28	lost three	positive
32	Germanium	Ge^{4+}	32	28	lost four	positive
33	Arsenic	As^{3-}	33	36	gained three	negative
34	Selenium	Se^{2-}	34	36	gained two	negative
35	Bromine	Br^-	35	36	gained one	negative
36	Krypton	Kr^{0}	36	36	lost none	neutral
37	Rubidium	Rb^+	37	36	lost one	positive
38	Strontium	Sr^{2+}	38	36	lost two	positive
39	Yttrium	Y^{3+}	39	36	lost three	positive
40	Zirconium	Zr^{4+}	40	36	lost four	positive
41	Niobium	Nb^{5+}	41	36	lost five	positive
42	Molybdenum	Mo^{6+}	42	36	lost six	positive
43	Technetium	Tc^{7+}	43	36	lost seven	positive
44	Ruthenium	Ru^{8+}	44	36	lost eight	positive
45	Rhodium	Rh^{9+}	45	36	lost nine	positive
46	Palladium	Pd^{10+}	46	36	lost ten	positive
47	Silver	Ag^+	47	46	lost one	positive
48	Cadmium	Cd^{2+}	48	46	lost two	positive
49	Indium	In^{3+}	49	46	lost three	positive
50	Mercury	Hg^{2+}	80	78	lost two	positive
51	Thallium	Tl^+	81	80	lost one	positive
52	Lead	Pb^{2+}	82	80	lost two	positive
53	Bismuth	Bi^{3+}	83	80	lost three	positive
54	Polonium	Po^{4+}	84	80	lost four	positive
55	Astatine	At^{5+}	85	80	lost five	positive
56	Radon	Rn^{0}	86	86	lost none	neutral
57	Francium	Fr^+	87	86	lost one	positive
58	Radium	Ra^{2+}	88	86	lost two	positive
59	Actinium	Ac^{3+}	89	86	lost three	positive
60	Thorium	Th^{4+}	90	86	lost four	positive
61	Protactinium	Pa^{5+}	91	86	lost five	positive
62	Uranium	U^{6+}	92	86	lost six	positive
63	Neptunium	Np^{7+}	93	86	lost seven	positive
64	Plutonium	Pu^{8+}	94	86	lost eight	positive
65	Americium	Am^{3+}	95	92	lost three	positive
66	Curium	Cm^{3+}	96	93	lost three	positive
67	Berkelium	Bk^{3+}	97	94	lost three	positive
68	Californium	Cf^{3+}	98	95	lost three	positive
69	Einsteinium	Es^{3+}	99	96	lost three	positive
70	Fermium	Fm^{3+}	100	97	lost three	positive
71	Mendelevium	Md^{3+}	101	98	lost three	positive
72	Nobelium	No^{2+}	102	99	lost two	positive
73	Lanthanum	La^{3+}	57	54	lost three	positive
74	Cerium	Ce^{4+}	58	54	lost four	positive
75	Praseodymium	Pr^{3+}	59	56	lost three	positive
76	Neodymium	Nd^{3+}	60	57	lost three	positive
77	Europium	Eu^{3+}	63	60	lost three	positive
78	Gadolinium	Gd^{3+}	64	61	lost three	positive
79	Terbium	Tb^{3+}	65	62	lost three	positive
80	Dysprosium	Dy^{3+}	66	63	lost three	positive
81	Ytterbium	Yb^{2+}	70	68	lost two	positive
82	Lutetium	Lu^{3+}	71	68	lost three	positive
83	Hafnium	Hf^{4+}	72	68	lost four	positive
84	Tantalum	Ta^{5+}	73	68	lost five	positive
85	Tungsten	W^{6+}	74	68	lost six	positive
86	Rhenium	Re^{7+}	75	68	lost seven	positive
87	Osmium	Os^{8+}	76	68	lost eight	positive
88	Iridium	Ir^{9+}	77	68	lost nine	positive
89	Platinum	Pt^{2+}	78	76	lost two	positive
90	Gold	Au^+	79	78	lost one	positive
91	Mercury	Hg^{2+}	80	78	lost two	positive
92	Thallium	Tl^+	81	80	lost one	positive
93	Lead	Pb^{2+}	82	80	lost two	positive
94	Bismuth	Bi^{3+}	83	80	lost three	positive
95	Polonium	Po^{4+}	84	80	lost four	positive
96	Astatine	At^{5+}	85	80	lost five	positive
97	Radon	Rn^{0}	86	86	lost none	neutral
98	Francium	Fr^+	87	86	lost one	positive
99	Radium	Ra^{2+}	88	86	lost two	positive
100	Actinium	Ac^{3+}	89	86	lost three	positive

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