1. Find the Volume of a Drop of Solution

Mass of first five drops of water	O. 03g		
Average mass of one drop of water	.0060		
Mass of second five drops of water	0.049		
Average mass of one drop of water	-0020		
Mass of third five drops of water	0.000		
Average mass of one drop of water	0.08		
Average of masses of one drop of water	.0053		
Volume of one drop of water	5.3x10°		

2. Determine the Reaction Rate and Calculate the Rate Law

	Time (s)				Reaction	Initial Concentrations, M.		
	Trial	Trial 2	Average	Temp (°C)	Rate (AVs)	[I.]	[BrO ₃ -]	[H*]
Exp.	17.97	14.30	16.125	20.0	8.9016	-0017	.6067	.6167
Exp.	650	205	6.78	20.0	Z-05110-6	33ص	-0067	.6117
Ехр. Э	4.29	3.74	4.02	200	3.46×10 ⁻⁶	2017	.00200	.6117
Ехр. 4					2.8626		.0067	.0233
Exp. 5	4.91	_			3.44 x10+		-0100	.0250

Sample Calculations: Santanos = Santanos + S

Rate of Reaction	1.3dx10-5 -
Initial Concentration	.01M.2 = n = .0017M I

Iodine Clock Experiment Lab Report

United States. National Bureau of Standards

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