

## Periodic Table Named-Groups Worksheet – 2

**DIRECTIONS:** Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

The image shows a crossword puzzle grid with letters placed in specific cells. The letters are as follows:

- Row 1:** A (1,1), B (1,13)
- Row 2:** C (2,1), D (2,10), E (2,11)
- Row 3:** F (3,2), G (3,7), H (3,9), K (3,13)
- Row 4:** L (4,2), I (4,11)
- Row 5:** J (5,1), Z (5,3), M (5,7), N (5,9), O (5,11), P (5,13)
- Row 6:** Q (6,1), R (6,3), S (6,7), T (6,8), U (6,9)

Below the main grid, there is a separate row of cells. The first two cells are labeled 'V' and 'X'. Arrows point from the letter 'R' in the main grid to these two cells, indicating the start of two words.

The remaining cells in this row are empty, except for the letters W, Y, and a blank space at the end.

List elements in alphabetical order when more than 1 element answers a question.

- \_\_\_\_\_ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid
- \_\_\_\_\_ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "E" **C)** "K" **D)** A, B, and C **E)** A and B only
- \_\_\_\_\_ 3) What is the letter of the element whose atomic number is closest to the element labeled "E"?
- \_\_\_\_\_ 4) Which letter labels the halogen element with the lowest atomic number?
- \_\_\_\_\_ 5) Which of these elements is the least reactive? **A)** "D" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"
- \_\_\_\_\_ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "K" **D)** "N" **E)** "Y"

continued on other side

# Holt Science And Technology Periodic Table Answers

**Ralph Thomas Myers**



## **Holt Science And Technology Periodic Table Answers:**

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