Chemistry of Life

# Study Guide A

#### **Answer Key**

## SECTION 1. ATOMS, IONS, AND MOLECULES

- 1. nucleus: dense center of an atom
- 2. neutron: particle with no electrical charge
- protonc particle with positive electrical charge
- electron: particle with negative electrical charge
- 5. compounds
- 6. elements
- 7. follow
- 8. false
- 9. time
- 10, trice
- 11. outermost
- 12. strong
- 13. electrons
- 14. covalent
- 15. compound
- 16. element
- 17, jonic bond
- 18, covalent bond
- 19. annexes
- 20. molecule

#### SECTION 2. PROPERTIES OF WATER

- I. false.
- 2. true
- 3. true
- 4, 21
- 5. b.
- 6. c.
- 7. evenly
- S. solvent
- 9. nonpolar
- more acidic, neutral, more basic.
- 11. solute

## SECTION 3. CARBON-BASED MOLECULES

- I. true
- 2. fallow
- 3. bruse
- 4. fished
- Students should sketch one of the following, based on Figure 3.1 in the student text: straight chain, branched chain, or ring.
- 6. Provide energy
- 7. starches, sugar
- 8. Store energy
- 9. fat, oils
- 10, source of amino acids
- 11. beans, meat, nuts
- 12. map for making proteins
- 13. DNA, RNA
- 14. polymer

#### SECTION 4. CHEMICAL REACTIONS

- 1. reactants, products; reactants, products
- 2. chemical bonds
- 3. reactants
- 4. adoress
- 5. same rate
- 6. false
- 7. true
- S. true 9. false
- 10, false
- chemical reaction that absorbs more energy than it releases
- chemical reaction that releases more energy than it absorbs
- amount of energy that needs to be absorbed for a chemical reaction to start
- substances changed during a chemical reaction
- 15. substances made by a chemical reaction
- state reached when reactants and products are made at the same rate
- amount of energy that will break a bond between two atoms

# **Holt Science And Technology Chemical Bonding 65**

**Julia Schneider** 

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