

Section 1: Chemistry

SECTION 1

Answer each question on the following questions in the space provided.

1. _____ a. The following are chemical equations representing the reaction of hydrogen gas with oxygen gas.
 - (a) $2H_2 + O_2 \rightarrow 2H_2O$
 - (b) $2H_2 + O_2 \rightarrow 2H_2O$
 - (c) $2H_2 + O_2 \rightarrow 2H_2O$
 - (d) $2H_2 + O_2 \rightarrow 2H_2O$
2. _____ b. Hydrogen gas is a colorless, odorless, and tasteless gas. It is the most abundant element in the universe.
 - (a) It is the most abundant element in the universe.
 - (b) It is the most abundant element in the universe.
 - (c) It is the most abundant element in the universe.
 - (d) It is the most abundant element in the universe.
3. _____ c. Hydrogen gas is a colorless, odorless, and tasteless gas. It is the most abundant element in the universe.
 - (a) It is the most abundant element in the universe.
 - (b) It is the most abundant element in the universe.
 - (c) It is the most abundant element in the universe.
 - (d) It is the most abundant element in the universe.
4. _____ d. Hydrogen gas is a colorless, odorless, and tasteless gas. It is the most abundant element in the universe.
 - (a) It is the most abundant element in the universe.
 - (b) It is the most abundant element in the universe.
 - (c) It is the most abundant element in the universe.
 - (d) It is the most abundant element in the universe.

5. Write the balanced chemical equation for the following reaction:



a. Balance the equation:



b. Write the balanced chemical equation for the reaction:



c. Write the balanced chemical equation for the reaction:



Holt Stoichiometry Section 1 Review Answers

Lauren Gardner



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