



Example 5

Draw the Lewis diagram for each molecule.

1. N_2
2. CH_2O (The carbon atom is the central atom.)

Solution

1. The bond between the two nitrogen atoms is a triple bond. The Lewis diagram for N_2 is as follows:



2. In CH_2O , the central atom is surrounded by two different types of atoms. The Lewis diagram that fills each atom's valence electron shell is as follows:



Note

One application of CH_2O , also called formaldehyde, is the preservation of biological specimens. Aqueous solutions of CH_2O are called formalin and have a sharp, characteristic (pungent) odor.

Skill-Building Exercise

Draw the Lewis diagram for each molecule.

1. O_3
2. C_2H_2

Concept Review Exercise

1. What is one clue that a molecule has a multiple bond?

Answer

1. If single bonds between all atoms do not give all atoms (except hydrogen) an octet, multiple covalent bonds may be present.

Key Takeaway

Holt Science Biology Ecosystem Answer Key

Richard Bailey



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