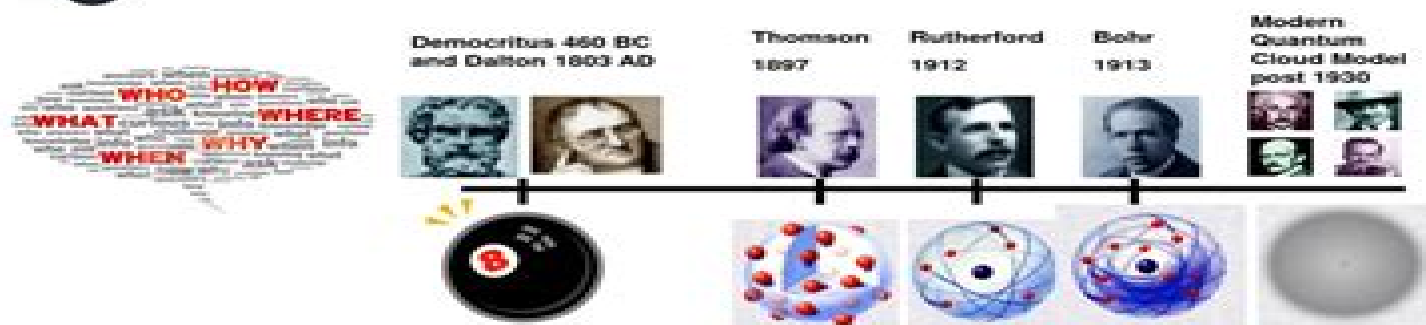




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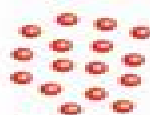


John Dalton (1766 – 1844):

John Dalton was an English chemist. His ideas form the atomic theory of matter. Here are his ideas.

- All elements are composed (made up) of atoms. It is impossible to divide or destroy an atom.
- All atoms of the same elements are alike. (One atom of oxygen is like another atom of oxygen.)
- Atoms of different elements are different. (An atom of oxygen is different from an atom of hydrogen.)
- Atoms of different elements combine to form a compound. These atoms have to be in definite whole number ratios. For example, water is a compound made up of 2 atoms of hydrogen and 1 atom of oxygen (a ratio of 2:1). Three atoms of hydrogen and 2 atoms of oxygen cannot combine to make water.

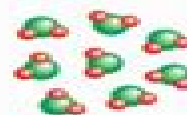
Dalton's Atomic Theory



Atoms of element X



Atoms of element Y



Compound of elements X and Y

1. What is the name of John Dalton's theory? **THE ATOMIC THEORY OF MATTER**
2. What are elements made of? **ELEMENTS ARE MADE OF ATOMS**
3. An atom of hydrogen and an atom of carbon are **DIFFERENT**
4. What are compounds made of? **COMPOUNDS ARE MADE OF ATOMS OF TWO OR MORE DIFFERENT ELEMENTS**
5. The ratio of atoms in HCl is: a) 1:3 b) 2:1 c) **1:1**
6. The ratio of atoms in H₂O is: a) 1:3 b) **2:1** c) 1:1 **2 atoms of hydrogen and 1 atom of oxygen**

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Table of Contents History Of The Atom Worksheet Answers

1. Understanding the eBook History Of The Atom Worksheet Answers
 - The Rise of Digital Reading History Of The Atom Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying History Of The Atom Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an History Of The Atom Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from History Of The Atom Worksheet Answers
 - Personalized Recommendations
 - History Of The Atom Worksheet Answers User Reviews and Ratings
 - History Of The Atom Worksheet Answers and Bestseller Lists

5. Accessing History Of The Atom Worksheet Answers Free and Paid eBooks
 - History Of The Atom Worksheet Answers Public Domain eBooks
 - History Of The Atom Worksheet Answers eBook Subscription Services
 - History Of The Atom Worksheet Answers Budget-Friendly Options
6. Navigating History Of The Atom Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - History Of The Atom Worksheet Answers Compatibility with Devices
 - History Of The Atom Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of History Of The Atom Worksheet Answers
 - Highlighting and Note-Taking History Of The Atom Worksheet Answers
 - Interactive Elements History Of The Atom Worksheet Answers
8. Staying Engaged with History Of The Atom Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers History Of The Atom Worksheet Answers
9. Balancing eBooks and Physical Books History Of The Atom Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection History Of The Atom Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine History Of The Atom Worksheet Answers
 - Setting Reading Goals History Of The Atom Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of History Of The Atom Worksheet Answers
 - Fact-Checking eBook Content of History Of The Atom Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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