

Name Lia Kerner

GRAM FORMULA MASS

Determine the gram formula mass (the mass of one mole) of each compound below.

1. KMnO_4 158 g
 $39 + 55 + 4(16) = 158$

2. KCl 74 g
 $39 + 35 = 74$

3. Na_2SO_4 142 g
 $2(23) + 32 + 4(16) = 142$

4. $\text{Ca(NO}_3)_2$ 164 g
 $40 + 2(14 + 3 \times 16) = 164$

5. $\text{Al}_2(\text{SO}_4)_3$ 342 g
 $2(27) + 3(32 + 4 \times 16) = 342$

6. $(\text{NH}_4)_3\text{PO}_4$ 149 g
 $3(14 + 4 \times 1) + 31 + 4 \times 16 = 149$

7. $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ 250 g
 $64 + 96 + 5(2 + 16) = 250$

8. $\text{Mg}_3(\text{PO}_4)_2$ 264 g
 $3(24) + (96) \times 2 = 264$

9. $7\text{H}_2(\text{C}_2\text{H}_3\text{O}_2) \cdot 2\text{H}_2\text{O}$ 219 g

Gram Formula Mass Answers

SJ Ball



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