

Compounds

Question 1: Choose the correct letter for each question:

- 1.1 Group I is known as
 A transition metals
 B alkali earth metals
 C alkali metals
 D earth metals
- 1.2 Group II is known as
 A transition metals
 B alkaline earth metals
 C alkali metals
 D earth metals
- 1.3 Group VII is known as
 A noble gases
 B halogens
 C halogen gases
 D noble gases
- 1.4 Group VIII is known as
 A non-metals
 B noble gases
 C halogen gases
 D noble gases
- 1.5 Which of the following elements has one electron in its outer orbital?
 A Phosphorous
 B Gallium
 C Strontium
 D Barium
- 1.6 Which element is the odd one out according to the periodic table?
 A Nitrogen
 B Antimony
 C Arsenic
 D Bismuth
- 1.7 Identify the element in group 13 and period 5
 A Germanium
 B Aluminium
 C Tin
 D Arsenic
- 1.8 Identify the element in group 14 and period 4
 A Calcium
 B Strontium
 C Silicon
 D Silicon
- 1.9 Which of the following is an example of a chemical property of elements?
 A Conductivity
 B Density
 C Heat capacity
 D Reactivity

Grade9 Naturalscience Exemplar

R Pring



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