

ঘের ও কীটপতঙ্গের অস্বাস্থ্যের নির্বাহী পদ্ধতির প্রয়োজন

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{16.5} = 1.21$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

চলকের ভর, $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = ?$

উদাহরণে উল্লিখিত X-পদ্ধতিতে 10 mg এর মধ্যে কত পরিমাণ দ্রব্য X-পদ্ধতিতে 0.5 g 2% (W/V) HCl দ্রবণ দিয়ে কাজ করা হয়েছে।

যদিও X-পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও Y-পদ্ধতিটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব। নিম্নে তা বিশ্লেষণ করা হলো।

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে দূষণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে দূষণ ঘটে না।

$(\text{CN})_2 + \text{H}_2\text{O} \rightarrow \text{HCN} + \text{HCN}$

অতএব, কোনো কোনো ক্ষেত্রে কম কঠিন রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

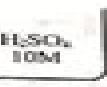
উদাহরণ



পাত্র-ক



পাত্র-খ



পাত্র-ঘ

কি. কে. ১০। এর ক. ১/

১. উদাহরণে দ্রব্য কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঙ্গার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অস্বাস্থ্যের দ্রবণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪

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১. বিশ্লেষণের নিমিত্ত বীমের উপর রাইডার স্থাপন করলে বীমের প্রতি স্কেলের জন্য যে ছবি পাওয়া যায়, তাকে রাইডার ছবি বলা হয়।

২. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় প্রায় 100 ml রাসায়নিক উপাদান সন্নিবেশিত ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে দূষণ উল্লেখযোগ্যভাবে কমে আসে।

৩. অতএব, কোনো কোনো ক্ষেত্রে কম কঠিন রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশ বান্দব।

৪. উদাহরণের দ্রবণটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M দ্রবণ। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M দ্রবণ হলে পাত্র H_2SO_4 দ্রবণ। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঙ্গার প্রতীকটি অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঙ্গার প্রতীক নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রতীক	ক্ষয়কারী (Corrosive)
আমদান	ক্ষয়কারী রাসায়নিক উপাদান কলমে যে ক্ষয়কারী রাসায়নিক দ্রব্যকে বুঝায় তার সংস্পর্শে ত্বক ও অন্যান্য পদার্থের ক্ষয়কারী ক্ষতি সঞ্চিত হয়। এতে ত্বক পুড়ে যায় এবং অন্যান্য পদার্থ গলে যায়।
উদাহরণ	পাত্র কঠিন এসিডসলফুরিক (HCl , HNO_3 , H_2SO_4), পাত্র ক্ষয় (NaOH , KOH), H_2O_2 , অক্সিজেন লিকার প্রভৃতি দ্রবণ ইত্যাদি।
সংরক্ষণ পদ্ধতি	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র ক্ষয়কে, ক্ষয়কারী পদার্থের পিছনে রাখতে হবে।

উদাহরণের দ্রবণটি হলো 10M H_2SO_4 দ্রবণ এবং এর থেকে 1M ঘনমাত্রার দ্রবণ প্রস্তুত করতে হবে 1ম ও ২য় চলকের ব্যবহার করতে হবে।

1ম চলকের দ্রবণটি হলো 100 ml অক্সিজেনমিত্রিক দ্রবণ ও এর দ্রবণটি পিঁপেট। সেমি-মাইক্রো অক্সিজেনমিত্রিক দ্রবণের আয়তন 100 ml। তাই আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রবণ তৈরি করতে হবে।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 দ্রবণের সঙ্গে পানি যোগ করতে হবে (100-10) mL = 90 mL।

অতএব, পিঁপেটের সাহায্যে 10 ml প্রথম ঘনমাত্রার দ্রবণ নিয়ে তার সাথে অক্সিজেনমিত্রিক দ্রবণে 90 ml পানি যোগ করলে আমাদের কাঙ্ক্ষিত ঘনমাত্রার দ্রবণ পাওয়া যাবে।

উদাহরণ



চিত্র-A



চিত্র-B



চিত্র-C

কি. কে. ১০। এর ক. ১/

১. অক্সিজেনমিত্রিক কলমে কলম?
২. সাদা ছবি K, এর মান কখনো অসীম হয় না-ব্যাখ্যা করে।
৩. A চিত্রের হ্যাঙ্গারের উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক দ্রবণের ব্যবহারকালে খাখা কুঁচি দ্রবণে কঠিন উপাদান সম্পর্কে বিশ্লেষণ করে।

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১. অক্সিজেনমিত্রিক কলম হলো রাসায়নিক বা কৃত্রিম দ্রব্য সংরক্ষণ বা মুদ্রিত বা উপাদানসমূহকে জারণ প্রতিরোধ করে রাখার পদ্ধতি।

২. সামান্য K, এর মান কখনো অসীম বা শূন্য হতে পারে না। তাইটি ব্যাখ্যা করার জন্য নিম্নের উত্তরদ্রব্য মাইক্রোপদ্ধতি বিশ্লেষণ করি—



Hsc Chemistry 1 First Chapter Notes

WJ Hussar



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