

### Atomic Structure Answer Key

1. A negatively-charged particle moving around the nucleus of an atom is a(n) \_\_\_\_\_.
  - a. compound
  - b. electron**
  - c. atom
3. Why does an atom have no net electric charge?
  - a. Its subatomic particles carry no electrical charge.
  - b. The positively charged protons cancel out the negatively charged neutrons.
  - c. The positively charged neutrons cancel out the negatively charged electrons.
  - d. The positively charged protons cancel out the negatively charged electrons.**
5. Compared to the charge and mass of a proton, an electron has
  - a. the same charge and a smaller mass.
  - b. the same charge and the same mass.
  - c. opposite charge and smaller mass.**
  - d. opposite charge and the same mass.
7. The electron cloud is a visual model of the probable locations of electrons in an atom.
  - a. True**
  - b. False
9. In Bohr-Rutherford diagrams, neutrons are typically located in the rings around the nucleus.
  - a. True
  - b. False**
2. A positively-charged particle in the nucleus of an atom is a(n) \_\_\_\_\_.
  - a. electron
  - b. proton**
  - c. element
4. What is an isotope?
  - a. An element with equal number of protons and neutrons
  - b. An elements with differing numbers of protons and electrons
  - c. An element with differing number of protons and neutrons**
  - d. An element with equal number of protons and electrons
6. A proton has approximately the same mass as \_\_\_\_\_.
  - a. a neutron**
  - b. a beta particle
  - c. an alpha particle
  - d. an electron
8. What is the nucleus made of?
  - a. protons
  - b. electrons
  - c. neutrons
  - d. protons and electrons
  - e. protons and neutrons**
  - f. electrons and neutrons
10. If atoms of a halogen nonmetal (Group 17) gain one electron, the atoms then have \_\_\_\_\_.
  - a. no valence electrons
  - b. 7 valence electrons
  - c. 8 valence electrons**
  - d. 17 valence electrons

# Georgia Science Grade 8 Coach Answer Key

**Ian Pickup**



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