

# Concept Review

## Section: Families of Elements

1. **Classify** each of the following elements as an alkali metal, alkaline-earth metal, transition metal, or semiconductor based on its position in the periodic table.

\_\_\_\_\_ **a.** rubidium, Rb

\_\_\_\_\_ **b.** silicon, Si

\_\_\_\_\_ **c.** silver, Ag

\_\_\_\_\_ **d.** barium, Ba

2. **Classify** each of the following elements as a halogen, noble gas, or other nonmetal based on its position in the periodic table.

\_\_\_\_\_ **a.** carbon, C

\_\_\_\_\_ **b.** chlorine, Cl

\_\_\_\_\_ **c.** radon, Rn

\_\_\_\_\_ **d.** phosphorus, P

3. **Predict** which of the following ions would be likely to form:

\_\_\_\_\_ **a.**  $\text{Na}^{2+}$

\_\_\_\_\_ **d.**  $\text{Br}^-$

\_\_\_\_\_ **b.**  $\text{Cl}^+$

\_\_\_\_\_ **e.**  $\text{Ne}^-$

\_\_\_\_\_ **c.**  $\text{Ca}^{2+}$

\_\_\_\_\_ **f.**  $\text{Ne}^+$

4. **Explain** why chlorine, Cl, is very reactive, whereas argon, Ar, is unreactive.

5. **Analyze** the following pairs of elements, and determine whether each pair has similar or different reactivities.

\_\_\_\_\_ **a.** potassium, K, and rubidium, Rb

\_\_\_\_\_ **b.** calcium, Ca, and barium, Ba

\_\_\_\_\_ **c.** sodium, Na, and chlorine, Cl

\_\_\_\_\_ **d.** helium, He, and krypton, Kr

# Fusion Fission Worksheet Answers

**Clifford Lane**



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