

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$\text{মলের ভর, } m = 20 \text{ g}$$

$$\text{HCl এর আপেক্ষিক ভর,}$$

$$M = 36.5 \text{ g/mol}$$

$$\text{চলকের আয়তন, } V = 1 \text{ L}$$

$$\text{চলকের ঘনমাত্রা } S = ?$$

উদাহরণে দ্রবীভূত X-পদার্থিতে 10 mg এর মধ্যে কয় পরিমাণ দ্রব
এ X-পদার্থিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl দ্রব দিতে কাজ করা হয়েছে।

যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও
Y-পদার্থটি হলো ছোটো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির
ছোটো সেমি-মাইক্রো পদ্ধতি পরিবেশে ব্যবহৃত। নিম্নে তা বিশ্লেষণ করা
হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
নিয়ন্ত্রিত বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05
থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশে দূষণ উল্লেখযোগ্যভাবে কমে আসে।
এছাড়া সেমি-মাইক্রো পদ্ধতিতে H₂S এর পরিমাণে
কোয়ালিটাইটিভ ব্যবহার করা হয় ফলে পরিবেশে দূষণ ঘটে না।



অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশে দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশে ব্যবহৃত।

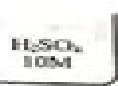
উদাহরণ



পাত্র-ক



পাত্র-খ



পাত্র-গ

১. টাইট্রার দ্রবক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশে ব্যবহৃত কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রক্রিয়ায় অঙ্কন করা এবং
ব্যাখ্যা করা।
৪. উদাহরণের অংশের দ্রবণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার
করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করা। ৪
ও না প্রদেয় উত্তর

৫. বিশ্লেষণের নিমিত্ত বীমের উপর টাইট্রার স্থাপন করলে বীমের প্রতি
দৃশ্যের অন্য যে দর পাওয়া যায়, তাকে টাইট্রার দ্রবক বলে।

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
নিয়ন্ত্রিত বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ
(0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
কাজ করা হয়। ফলে পরিবেশে দূষণ উল্লেখযোগ্যভাবে কমে আসে।
অতএব, কোনো কোনো ক্ষেত্রে কম কষ্টের রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশে দূষণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশে ব্যবহৃত।

উদাহরণের দ্রবণটি হলো সালফিউরিক এসিড (H₂SO₄) এর 10 M
দ্রব। H₂SO₄ একটি জারক ও ক্ষয়কারী পদার্থ। H₂SO₄ এর 10 M
দ্রব যখন পাত্র H₂SO₄ দ্রবণ। পাত্র H₂SO₄ পদার্থকে ক্ষয়কারী হ্যাঞ্জার
এইভাবে অঙ্কন করা হয়। পাত্র H₂SO₄ এর জন্য ক্ষয়কারী হ্যাঞ্জার
এইক নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রক্রিয়া	ক্ষয়কারী (Corrosive)
আমদান	ক্ষয়কারী রাসায়নিক উপাদান কলমে যে ক্ষয় রাসায়নিক দ্রব্যকে খুবই দ্রুত ক্ষয় সম্পর্কে তুলে ও অন্য পদার্থের ক্ষয়কারী ক্ষতি সঞ্চিত হয়। এতে তুলে পুড়ে যায় এবং অন্যদিকে পদার্থ পুড়ে যায়।
উদাহরণ	পাত্র কঠিন এসিডসমূহ (HCl, HNO ₃ , H ₂ SO ₄) পাত্র ক্ষয় (NaOH, KOH), H ₂ O ₂ , অক্সিজেন লিকার ক্রিডিং দ্রবণ ইত্যাদি।
সংরক্ষণ পদ্ধতি	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র ক্ষয়কে, ক্ষয় সম্পর্কে বিপাকের পোতলে রাখতে হবে।

উদাহরণের দ্রবণটি হলো 10M H₂SO₄ দ্রবণ এবং এর থেকে 1M
ঘনমাত্রার দ্রবণ প্রস্তুত করতে হবে 1ম ও ১ম চলকের ব্যবহার করতে
হবে।

1ম চলকের দ্রবণটি হলো 100 ml অক্সিজেনমুক্ত জল ও এর দ্রব
নিম্নে। যেহেতু অক্সিজেনমুক্ত জলের অঙ্কন 100 ml, তাই
আমাদের 100 ml 1M ঘনমাত্রার H₂SO₄ দ্রবণ তৈরি করতে হবে।
দ্রবণের অনুপাত দ্রব হতে জানতে পারি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H₂SO₄ দ্রবণের সঙ্গে পানি যোগ করতে হবে
(100-10) mL = 90 mL
অতএব, পাত্রের দ্রবণে 10 ml দ্রব ঘনমাত্রার দ্রবণ নিয়ে তার
সঙ্গে অক্সিজেনমুক্ত জলকে 90 ml পানি যোগ করলে আমাদের
কামিষ্ঠ ঘনমাত্রার দ্রবণ পাওয়া যাবে।

উদাহরণ



চিত্র-A



চিত্র-B



চিত্র-C

১. অক্সিজেনমুক্ত জলকে বলা।
২. সামান্য দ্রবক K₂ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা
করা।
৩. A চিত্রের হ্যাঞ্জারে উত্তর করার কৌশল বর্ণনা করা।
৪. B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক দ্রবণের
ব্যবহারকালে খাখা কৃষ্টি দ্রবণে কবীর উপর সম্পর্কে
বিশ্লেষণ করা। ৪
ও না প্রদেয় উত্তর

অক্সিজেনমুক্ত জল এক ধরনের রাসায়নিক বা কৃত্রিম দ্রবণ সঞ্চিত
বা দ্রব বা উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে তুলে করে
খাখার সঞ্চিত করে।

সামান্য K₂ এর ঘন কখনো অসীম বা শূন্য হতে পারে না।
উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণ মাইক্রোপদ্ধতি বিশ্লেষণ করি-
 $A + B \longrightarrow C + D$

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