

UNIT 2 PART 1 STUDY GUIDE

READING COMPREHENSION

1. **What's matter?**
Matter makes up everything except energy (heat and light)
2. **Give three examples of matter.**
Solid (steel), liquid (water), and gas (helium) except light is energy
3. **Discuss the difference between a physical property and a chemical property.**
Physical properties can be observed using five senses without changing the substance.
Chemical property is observed when the substance changes.
4. **Give examples of physical properties.**
Color, texture, temperature, shape, size, mass, conductivity, density, magnetism
5. **Give examples of chemical property.**
Flammability, ability to react with other substances, reactivity, toxicity
6. **Discuss the difference between solubility property and a composition property.**
Solubility property shows how well one thing dissolves into another, and is a composition property because it's not changed by physical changes. Type is constant
7. **What's an element? Give an example of an element.**
It is the smallest chemical matter that cannot be broken down by physical changes.
Gold
8. **How is an element different from a compound? Give an example of a compound.**
An element cannot be broken down, a compound can be broken down into elements without losing identity.
Sodium chloride
9. **What's a mixture?**
A mixture is a combination of substances where the substances do not give up their properties. (homogeneous mixtures)
Air
10. **How like is homogeneous mixture. Give an example of a homogeneous mixture.**
Homogeneous mixtures are uniform throughout.
Salt water
11. **How like is heterogeneous mixture. Give an example of a heterogeneous mixture.**
Heterogeneous mixtures are not uniform throughout. They have the individual parts that make them.
12. **How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?**
You can see the individual substances of the heterogeneous mixture while the homogeneous mixture looks the same throughout. An example of a heterogeneous mixture is a salad.

What's up? What's not

1. **What are the three states of matter?**

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Randal S. Chase



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