

Name: _____ Date: _____ Per: _____

Ionic Bonding WebQuest

Part 1: Go to http://www.visionlearning.com/library/module_viewer.php?mid=55

Read the **Chemical Bonding** and **Ionic Bonding** sections, and answer the following questions:

1. Elements react with each other to form _____ and _____.
2. What two elements is table salt composed of? _____ and _____.
3. Who contributed to our knowledge of valence electrons? _____.
4. What are the two types of bonding? _____ and _____.
5. Briefly describe how an ionic bond forms: _____.

6. Look at the picture of the **NaCl Crystal Schematic**. Explain why there is no such thing as a single molecule of salt: _____.

7. Click the link [The reaction of sodium with chlorine](#) and watch through the video there. Briefly describe each of the steps involved with the reaction of sodium with chlorine: _____.

Part 2: Watch the following clips on Ionic Bonding:

- <http://search2video.net/ionic-bonding-video-ofRtGtkaGwKP.html>
- <http://www.youtube.com/watch?v=QzjzDyzf8w0> (ignore the covalent part)
- http://www.mhhe.com/physac/chemistry/animations/chem_7a_ssp/comp1a2_11.swf (you can click on the NaF ionic compound and see an animation on how it forms.)

Part 3: Explore the link below. (You can click on the elements and compounds in the video for extra info.)
http://content.tutorvista.com/chemistry_10/content/media/048_electro_bonding.swf

8. Describe how ionic compounds form: _____.

Part 4: Take the quiz at the following website. Record your score here: _____ of 10.
http://www.softschools.com/quizzes/chemistry/ionic_bonding_formation/quiz993.html

Part 5: Dissociation of Salt (an ionic compound). Watch the following videos:

- <http://www.youtube.com/watch?v=EBtGzTAuF4o>
- <http://www.mhhe.com/physac/chemistry/essentialchemistry/flash/molvie1.swf>

9. Go to <http://phet.colorado.edu/en/simulation/sugar-and-salt-solutions>. Use the simulation and explain how salt and sugar dissolve in water: _____.

Ionic Covalent Bonding Webquest Answer Key

Rachel Turoscy



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