

What is the gas pressure inside the box shown in the figure?



$$P_{\text{gas}} = 1 \text{ atm} + \rho g h$$

$h = +$ right higher
 $h = -$ left higher

Mastering Physics

13.16

$$(13,600 \times 9.8 \times -0.1) = -0.1 \text{ m}$$

$$\rho = 13,600 \text{ kg/m}^3$$

$$-87,972 \text{ Pa}$$

$$8.8 \times 10^4 \text{ Pa}$$

1 atm	101,300 Pa
1 atm	



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Rosina Ehmann



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