

KEY

## Ionic Formula Practice

| Name                     | Cation             | Anion                                 | Formula                                       |
|--------------------------|--------------------|---------------------------------------|-----------------------------------------------|
| 1. calcium bromide       | $\text{Ca}^{+2}$   | $\text{Br}^{-1}$                      | $\text{Ca Br}_2$                              |
| 2. magnesium phosphide   | $\text{Mg}^{+2}$   | $\text{P}^{-3}$                       | $\text{Mg}_3 \text{P}_2$                      |
| 3. aluminum oxide        | $\text{Al}^{+3}$   | $\text{O}^{-2}$                       | $\text{Al}_2 \text{O}_3$                      |
| 4. barium sulfide        | $\text{Ba}^{+2}$   | $\text{S}^{-2}$                       | $\text{Ba S}$                                 |
| 5. sodium phosphide      | $\text{Na}^{+1}$   | $\text{P}^{-3}$                       | $\text{Na}_3 \text{P}$                        |
| 6. strontium chloride    | $\text{Sr}^{+2}$   | $\text{Cl}^{-1}$                      | $\text{Sr Cl}_2$                              |
| 7. calcium nitride       | $\text{Ca}^{+2}$   | $\text{N}^{-3}$                       | $\text{Ca}_3 \text{N}_2$                      |
| 8. ammonium bromide      | $\text{NH}_4^{+1}$ | $\text{Br}^{-1}$                      | $\text{NH}_4 \text{Br}$                       |
| 9. calcium sulfate       | $\text{Ca}^{+2}$   | $\text{SO}_4^{-2}$                    | $\text{Ca SO}_4$                              |
| 10. magnesium nitrate    | $\text{Mg}^{+2}$   | $\text{NO}_3^{-1}$                    | $\text{Mg}(\text{NO}_3)_2$                    |
| 11. sodium carbonate     | $\text{Na}^{+1}$   | $\text{CO}_3^{-2}$                    | $\text{Na}_2 \text{CO}_3$                     |
| 12. zinc hydroxide       | $\text{Zn}^{+2}$   | $\text{OH}^{-1}$                      | $\text{Zn}(\text{OH})_2$                      |
| 13. barium nitrite       | $\text{Ba}^{+2}$   | $\text{NO}_2^{-1}$                    | $\text{Ba}(\text{NO}_2)_2$                    |
| 14. aluminum sulfite     | $\text{Al}^{+3}$   | $\text{SO}_3^{-2}$                    | $\text{Al}_2(\text{SO}_3)_3$                  |
| 15. lead II nitrate      | $\text{Pb}^{+2}$   | $\text{NO}_3^{-1}$                    | $\text{Pb}(\text{NO}_3)_2$                    |
| 16. copper I phosphate   | $\text{Cu}^{+1}$   | $\text{PO}_4^{-3}$                    | $\text{Cu}_3 \text{PO}_4$                     |
| 17. ammonium carbonate   | $\text{NH}_4^{+1}$ | $\text{CO}_3^{-2}$                    | $(\text{NH}_4)_2 \text{CO}_3$                 |
| 18. tin IV nitrate       | $\text{Sn}^{+4}$   | $\text{NO}_3^{-1}$                    | $\text{Sn}(\text{NO}_3)_4$                    |
| 19. nickel II acetate    | $\text{Ni}^{+2}$   | $\text{C}_2\text{H}_3\text{O}_2^{-1}$ | $\text{Ni}(\text{C}_2\text{H}_3\text{O}_2)_2$ |
| 20. chromium III sulfate | $\text{Cr}^{+3}$   | $\text{SO}_4^{-2}$                    | $\text{Cr}_2(\text{SO}_4)_3$                  |
| 21. iron III nitrite     | $\text{Fe}^{+3}$   | $\text{NO}_2^{-1}$                    | $\text{Fe}(\text{NO}_2)_3$                    |
| 22. cobalt II hydroxide  | $\text{Co}^{+2}$   | $\text{OH}^{-1}$                      | $\text{Co}(\text{OH})_2$                      |

# Ionic Formula Writing Kit Lab Answer Key

**N Noddings**



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web film craft türkçesi film craft nedir sinema sinemacılık film craft ingilizcede ne demek film craft nerede nasıl kullanılır film boş film Çevirmek film kaplamak yaygı Çok ince ve yeğni kabuk ya da katman Çevirim eylemi