

Name Solutions

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Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers 1ve ⁻ to F	Li^+
F	7 ve	accepts 1ve ⁻ from Li	F^-

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $Li + Cl \Rightarrow LiCl$ 	Li	1 ve	donates 1e ⁻	Li^+
	Cl	1 ve	accepts 1e ⁻	Cl^-
2) $Ca + O \Rightarrow CaO$ 	Ca	2 ve	donates 2e ⁻	Ca^{+2}
	O	6 ve	accepts 2e ⁻	O^{-2}
3) $Be + F \Rightarrow BeF_2$ 	Be	2 ve	donates 2e ⁻ to each F	Be^{+2}
	F	7 ve	each F accepts 1ve each	F^-
4) $Mg + S \Rightarrow MgS$ 	Mg	2 ve	donates 2e ⁻ to S	Mg^{+2}
	S	6 ve	accepts 2e ⁻	S^{-2}
5) $K + F \Rightarrow KF$ 	K	1 ve	donates 1ve ⁻	K^+
	F	7 ve	accepts 1ve ⁻	F^-

Ionic Bonding Worksheet Answer Key

NARAYAN CHANGDER



Ionic Bonding Worksheet Answer Key:

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