

ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পদ্ধতির প্রয়োগের

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$\text{মলের ভর, } m = 20 \text{ g}$$

$$\text{HCl এর আপেক্ষিক ভর,}$$

$$M = 36.5 \text{ g/mol}$$

$$\text{চলকের আয়তন, } V = 1 \text{ L}$$

$$\text{চলকের ঘনমাত্রা } S = ?$$

উদাহরণে দ্রবীভূত X-পদার্থিতে 10 mg এর মধ্যে কত পরিমাণ দ্রব
এ X-পদার্থিতে 0.5 g 2% $\left(\frac{W}{V}\right)$ HCl দ্রবের সাথে মিশ্রিত করা হয়েছে।

যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও
V-পদ্ধতিতে যথেষ্ট ঘনমাত্রা আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির
মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশে ব্যবহৃত। নিম্নে তা বিশ্লেষণ করা
হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
একমাত্র বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05
থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
পরীক্ষা করা হয়। ফলে পরিবেশে দ্রবণ উল্লেখযোগ্যভাবে কমে আসে।
এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে
কোয়ান্টিটাইভ ব্যবহার করা হয় ফলে পরিবেশে দ্রবণ ঘটি না।



অতএব, কোনো কোনো ক্ষেত্রে কম কৃত্রিম রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশে দ্রবণের সন্নিবেশ কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশে ব্যবহৃত।

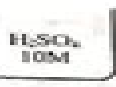
উদাহরণ



পাত্র-ক



পাত্র-খ



পাত্র-গ

১. টাইট্রার দ্রবক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশে ব্যবহৃত কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রক্রিয়াটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অংশের দ্রবণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার দ্রবণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪ ও না প্রদেয় উত্তর

৫. বিশ্লেষণের নিমিত্ত কীমের উপর টাইট্রার স্থাপন করলে কীমের প্রতি
দ্রবণের জন্য যে দর পাওয়া যায়, তাকে টাইট্রার দ্রবক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি।
এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও
একমাত্র বিশ্লেষণে ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ
(0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে
পরীক্ষা করা হয়। ফলে পরিবেশে দ্রবণ উল্লেখযোগ্যভাবে কমে আসে।
অতএব, কোনো কোনো ক্ষেত্রে কম কৃত্রিম রাসায়নিক দ্রব্য ব্যবহৃত হয়।
ফলে পরিবেশে দ্রবণের সন্নিবেশ কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি
পরিবেশে ব্যবহৃত।

৭. উদাহরণের দ্রবণটি হলো সালফিউরিক এসিড (H_2SO_4) এর 10 M
দ্রবণ। H_2SO_4 একটি জারক ও ক্ষয়কারী পদার্থ। H_2SO_4 এর 10 M
দ্রবণ যখন পাত্র H_2SO_4 দ্রবণ। পাত্র H_2SO_4 পদার্থকে ক্ষয়কারী হ্যাঞ্জার
একভাবে অঙ্কন করা হয়। পাত্র H_2SO_4 এর জন্য ক্ষয়কারী হ্যাঞ্জার
একটি নিয়ে অঙ্কন করে ব্যাখ্যা করা হলো:

প্রক্রিয়া	ক্ষয়কারী (Corrosive)
আয়তন	ক্ষয়কারী রাসায়নিক উপাদান কালে যে ক্ষয় রাসায়নিক দ্রব্যকে ক্ষয়কারী দ্রব সলিউশনে তুলে ও অন্যদিকে পাত্রের মাধ্যমে ক্ষয়কারী ক্ষয়কারী দ্রব। এতে তুলে পুড়ে যায় এবং অন্যদিকে পাত্রের পুড়ে যায়।
উদাহরণ	পাত্র কঠিন এসিডসলফিউরিক (HCl , HNO_3 , H_2SO_4), পাত্র ক্ষয় (NaOH , KOH , H_2O_2), অ্যাসেনিক এসিড কঠিন দ্রবণ ইত্যাদি।
সংরক্ষণ পাত্রের	- পাত্র এসিডকে পোতলে রাখতে হবে। - পাত্র ক্ষয়কারী, ক্ষয়কারী পাত্রের পাত্রের পোতলে রাখতে হবে।

৮. উদাহরণের দ্রবণটি হলো 10M H_2SO_4 দ্রবণ এবং এর থেকে 1M
ঘনমাত্রার দ্রবণ প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করে
হবে।

1ম চলকের দ্রবণটি হলো 100 ml অ্যাসেনিক এসিড ও এর টি হলো
নিম্নের। যেহেতু অ্যাসেনিক এসিডের ঘনমাত্রা 100 ml, তাই
আমাদের 100 ml 1M ঘনমাত্রার H_2SO_4 দ্রবণ তৈরি করতে হবে।
দ্রবণের সন্নিবেশ পুড়ে যাবে অতএব জমি।

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml H_2SO_4 দ্রবণের সঙ্গে পানি যোগ করতে হবে
(100-10) mL = 90 mL
অতএব, পাত্রের মাধ্যমে 10 ml দ্রবণ ঘনমাত্রার দ্রবণ নিয়ে তার
সঙ্গে অ্যাসেনিক এসিডের 90 ml পানি যোগ করলে আমাদের
কামিষ্ট ঘনমাত্রার দ্রবণ পাওয়া যাবে।

উদাহরণ



চিত্র-A



চিত্র-B



চিত্র-C

১. অ্যাসিডক্লোরাইট কাকে বলে?
২. সাদা দ্রবক K₂ এর ঘন কখনো অসীম হয় না-ব্যাখ্যা
করে।
৩. A চিত্রের হ্যাঞ্জার উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চিত্রের মাধ্যমে সম্পর্কিত রাসায়নিক দ্রবণের
ব্যবহারকালে খাম্বা কৃতি দ্রবণে কবীর উপর সম্পর্কে
বিশ্লেষণ করে। ৪ ও না প্রদেয় উত্তর

৫. অ্যাসিডক্লোরাইট এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সলিউশন
বা দ্রবণ বা উপাদানসমূহকে জারণ প্রক্রিয়ায় জাল থেকে তুলে
খাবার সলিউশন করে।

৬. সামান্য K₂ এর ঘন কখনো অসীম বা শূন্য হতে পারে না।
উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উত্তরদ্রব্যী মাইক্রোপদ্ধতি
বিশ্লেষণ করি-
 $A + B \longrightarrow C + D$

Hsc Chemistry 1 First Chapter Notes

JG Myers



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