

Chemistry Midterm Study Guide

Periodic Table of the Elements

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- Period** - is the horizontal row of elements in the periodic table
ex - Chlorine has an atomic number of 17, so it is in the 3rd period.
- Group** - is the vertical column of elements in the periodic table
ex - Fluorine has an atomic number of 9, so it is in the 17th group.
- Electronegativity** - the ability of an atom to attract electrons towards itself
ex - Fluorine has the highest electronegativity of all elements.
- Ionization Energy** - the energy required to remove an electron from an atom
ex - Helium has the highest ionization energy of all elements.

High School Chemistry Midterm Study Guide

DP Hallahan



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