

Determining Empirical Formulas

1) 75% carbon, 25% hydrogen	Answer:	CH_4
Work: $\text{C } 75\% \rightarrow 75\text{g C} \left \frac{1 \text{ mol}}{12.01\text{g C}} = 6.24 \text{ mol} / 6.24 = 1 \right.$ $\text{H } 25\% \rightarrow 25\text{g H} \left \frac{1 \text{ mol H}}{1.01\text{g H}} = 24.75 \text{ mol} / 6.24 = 4 \right.$		$\boxed{\text{CH}_4}$
2) 52.7% potassium, 47.3% chlorine	Answer:	KCl
Work: $\text{K } 52.7\% \rightarrow 52.7\text{g K} \left \frac{1 \text{ mol K}}{39.10\text{g K}} = 1.34 \text{ mol} / 1.33 = 1 \right.$ $\text{Cl } 47.3\% \rightarrow 47.3\text{g Cl} \left \frac{1 \text{ mol Cl}}{35.45\text{g Cl}} = 1.33 \text{ mol} / 1.33 = 1 \right.$		$\boxed{\text{KCl}}$
3) 22.1% aluminum, 25.4% phosphorous, 52.5 % oxygen	Answer:	AlPO_4
Work: $\text{Al } 22.1\% \rightarrow 22.1\text{g Al} \left \frac{1 \text{ mol Al}}{26.98\text{g Al}} = 0.819 \text{ mol} / 0.819 = 1 \right.$ $\text{P } 25.4\% \rightarrow 25.4\text{g P} \left \frac{1 \text{ mol P}}{30.97\text{g P}} = 0.820 \text{ mol} / 0.819 = 1 \right.$ $\text{O } 52.5\% \rightarrow 52.5\text{g O} \left \frac{1 \text{ mol O}}{16.00\text{g O}} = 3.28 \text{ mol} / 0.819 = 4 \right.$		$\boxed{\text{AlPO}_4}$
4) 13% magnesium, 87% bromine	Answer:	MgBr_2
Work: $13\text{g Mg} \left \frac{1 \text{ mol Mg}}{24.31\text{g Mg}} = 0.54 \text{ mol Mg} / 0.54 = 1 \right.$ $87\text{g Br} \left \frac{1 \text{ mol Br}}{79.90\text{g Br}} = 1.09 \text{ mol Br} / 0.54 = 2 \right.$		$\boxed{\text{MgBr}_2}$
5) 32.4% sodium, 22.5% sulfur, 45.1% oxygen	Answer:	Na_2SO_4
Work: $32.4\text{g Na} \left \frac{1 \text{ mol}}{22.99\text{g Na}} = 1.41 \text{ mol Na} / 0.70 = 2 \right.$ $22.5\text{g S} \left \frac{1 \text{ mol S}}{32.37\text{g S}} = 0.70 \text{ mol S} / 0.70 = 1 \right.$ $45.1\text{g O} \left \frac{1 \text{ mol O}}{16\text{g O}} = 2.82 \text{ mol O} / 0.70 = 4 \right.$		$\boxed{\text{Na}_2\text{SO}_4}$

Empirical Formula Chemistry If8766 Answers

SA Dillow



Empirical Formula Chemistry If8766 Answers:

This book delves into Empirical Formula Chemistry If8766 Answers. Empirical Formula Chemistry If8766 Answers is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Empirical Formula Chemistry If8766 Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Empirical Formula Chemistry If8766 Answers
- Chapter 2: Essential Elements of Empirical Formula Chemistry If8766 Answers
- Chapter 3: Empirical Formula Chemistry If8766 Answers in Everyday Life
- Chapter 4: Empirical Formula Chemistry If8766 Answers in Specific Contexts
- Chapter 5: Conclusion

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examine the hind legs and front legs of the frog

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