

## Lab 1- What is around us?

2/1/2024

### Focus Questions

- 1) What types of compounds do you know? Give a few examples.
  - Two examples of compounds are ionic and covalent bonds.
- 2) When naming different types of compounds do you need to follow different rules? Explain using examples.
  - For naming covalent bonds first write the names of both elements. Then change the ending of the second one to ide and add the necessary prefixes.
    - $\text{SiCl}_4 \rightarrow$  Silicon Tetrachloride
  - For ionic compounds, write the metal's name first and then the non-metal. Change non-metals ending to ide. Add roman numerals if necessary.
    - $\text{MgBr}_2 \rightarrow$  Magnesium Bromide
    - $\text{CuBr} \rightarrow$  Copper (I) Bromide
- 3) Does it matter in which order you combine your elements when writing chemical formulas for ionic compounds?
  - Yes it does. The cation is typically written first which is the atom which is losing an electron. This means the positive ion is written first and then the negative ion is. The polyatomic ion is always written last.
- 4) Which laws govern the formation of compounds?
  - The law of mass action.

### Post-Lab Assessment Questions (20 points):

1. (6 pts) For the following pairs of ions, write the formula of the compound that you would expect them to form:

- a. barium and cyanide  $\text{Ba}(\text{CN})_2$
- b. cobalt(III) and sulfur  $\text{Co}_2\text{S}_3$
- c. iron(III) and sulfate ~~FeSO<sub>4</sub>~~  $\text{Fe}_2(\text{SO}_4)_3$

2. (6 pts) Platinum is a transition metal and forms two different ions,  $\text{Pt}^{2+}$  and  $\text{Pt}^{4+}$ . Write the formulas for the compounds for each platinum ion with:

- a. chloride ions  $\text{PtCl}_2$ ,  $\text{PtCl}_4$
- b. carbonate ions  $\text{PtCO}_3$ ,  $\text{Pt}(\text{CO}_3)_2$
- c. nitrate ions  $\text{Pt}(\text{NO}_3)_2$ ,  $\text{Pt}(\text{NO}_3)_4$

3. (4 pts) Name the following compounds:

- a.  $\text{N}_2\text{O}_5$  Dinitrogen Pentaoxide
- b.  $\text{AlCl}_3$  ~~Aluminum Chloride~~ Aluminium Chloride
- c.  $\text{K}_3\text{PO}_4$  Tripotassium Phosphate
- d.  $\text{Mn}_2\text{O}_7$  Manganese (VII) Oxide

4. (4 pts) Write the formulas for the following compounds:

- a. Sulfur tetrachloride  $\text{SCl}_4$
- b. Bromine dioxide  $\text{BrO}_2$
- c. Nickel(II) nitrate  $\text{Ni}(\text{NO}_3)_2$
- d. Calcium sulfide  $\text{CaS}$

# Hunter Lab Manual Chem 106 Lab

**Léo Pomar**



## **Hunter Lab Manual Chem 106 Lab:**

**Chemistry in Context Chem 106** Mark A. Griep, 2009-12-01 *A Laboratory Manual of Analytical Methods of Protein Chemistry, Including Polypeptides* Peter Alexander, Richard Joseph Block, Harold Palmer Lundgren, 1960 *A Laboratory Manual of Analytical Methods of Protein Chemistry*, 1960 **A Laboratory Guide in Pharmacology** Torald Hermann Sollmann, 1917 **Chem 106 Chemistry in Context 2** Mark A. Griep, 2017-01-03 *A Digest of Investigations in the Teaching of Science* Francis Day Curtis, 1926 **A Digest of Investigations in the Teaching of Science in the Elementary and Secondary Schools** Francis Day Curtis, 1926 *A Manual of Laboratory Experiences in Cell Biology* C. Edward Gasque, Clayton Edward Gasque, 1989 **The Composition, Structure and Reactivity of Proteins** P. Alexander, R. J. Block, 2014-06-05 *The Composition Structure and Reactivity of Proteins* *Quick Guide to Psychiatric Emergencies* Kimberly D. Nordstrom, Michael P. Wilson, 2018-03-07 This volume provides an on the go guide to the most common behavioral emergencies a physician may encounter Each chapter represents a disease state or symptom cluster and concisely summarizes the disease state provides background symptoms and signs differential diagnoses and immediate and long term treatment options All chapters conclude with a diagnosis or treatment algorithm or another easy to use visual tool Chapters named after a specific disease state or symptom cluster arranged alphabetically for use in the field The text begins with chapters covering patient evaluation getting a good history suicide risk assessment physical exam and when and how to use studies Written by experts in psychiatry and emergency medicine this text is the first to consider both medical perspectives in a concise guide *Quick Guide to Psychiatric Emergencies* is an excellent resource for psychiatrists emergency medicine physicians residents nurses and other medical professionals that handle behavioral emergencies on a regular basis

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