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CHAPTER 1 INTRODUCTION

- 1.1 (a) Matter is anything that occupies space and has mass. (b) Mass is a measure of the quantity of matter in an object. (c) Weight is the force that gravity exerts on an object. (d) A substance is matter that has a definite or constant composition and distinct properties. (e) A mixture is a combination of two or more substances in which the substances retain their distinct identities.
- 1.2 The scientifically correct statement is, "The mass of the student is 56 kg."
- 1.3 Table salt dissolved in water is an example of a homogeneous mixture. Oil mixed with water is an example of a heterogeneous mixture.
- 1.4 A physical property is any property of a substance that can be observed without transforming the substance into some other substance. A chemical property is any property of a substance that cannot be studied without converting the substance into some other substance.
- 1.5 Density is an example of an intensive property. Mass is an example of an extensive property.
- 1.6 (a) An element is a substance that cannot be separated into simple substances by chemical means.
(b) A compound is a substance composed of atoms of two or more elements chemically united in fixed proportions.
- 1.7 (a) Chemical property. Oxygen gas is consumed in a combustion reaction; its composition and identity are changed.
(b) Chemical property. The fertilizer is consumed by the growing plants; it is turned into vegetable matter (different composition).

General Chemistry The Essential Concepts Solution Manual

JL Elias



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Malone,Theodore O. Dolter,2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter General,

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