

ACS Final Exam Practice 1

The exam is 70 questions in 110 minutes. You may use the equation sheet and periodic table but no other materials

Section 1 Foundations

PQ-1. The speed of light is  $3.0 \times 10^8 \text{ m}\cdot\text{s}^{-1}$ . What is this speed in  $\text{nm}\cdot\text{ms}^{-1}$ ?

- (A)  $3.0 \times 10^{-1} \text{ nm}\cdot\text{ms}^{-1}$       (B)  $3.0 \times 10^2 \text{ nm}\cdot\text{ms}^{-1}$       (C)  $3.0 \times 10^{14} \text{ nm}\cdot\text{ms}^{-1}$       (D)  $3.0 \times 10^{23} \text{ nm}\cdot\text{ms}^{-1}$

PQ-4. What is the correctly reported mass of water based on the data in the table?

Mass of beaker and water	29.62 g
Mass of beaker only	28.3220 g

- (A) 1.3 g      (B) 1.30 g      (C) 1.298 g      (D) 1.2980 g

PQ-5. What is the name of  $\text{Ti}_2(\text{PO}_4)_3$ ?

- (A) titanium phosphate      (B) titanium(III) phosphate  
(C) titanium(IV) phosphate      (D) titanium tetraphosphate

PQ-6. The formula of strontium hexafluorosilicate is  $\text{SrSiF}_6$ . What is the formula of aluminum hexafluorosilicate?

- (A)  $\text{AlSiF}_6$       (B)  $\text{Al}_2(\text{SiF}_6)_3$       (C)  $\text{Al}_2\text{SiF}_6$       (D)  $\text{Al}_2(\text{SiF}_6)_2$

PQ-8. What is the correct name for  $\text{N}_2\text{O}_5$ ?

- (A) nitrogen(II) oxide      (B) nitrogen(V) oxide  
(C) dinitrogen oxide      (D) dinitrogen pentoxide

PQ-12. If a palladium nanoparticle has a density of  $12.0 \text{ g}\cdot\text{cm}^{-3}$ , what is the mass of a nanoparticle with a volume of  $1.84 \times 10^{-23} \text{ cm}^3$ ?

- (A)  $2.21 \times 10^{-23} \text{ g}$       (B)  $1.63 \times 10^{-20} \text{ g}$       (C)  $6.52 \times 10^{-21} \text{ g}$       (D)  $1.53 \times 10^{-22} \text{ g}$

# General Chemistry Answer Key

**Rachel Turoscy**



## **General Chemistry Answer Key:**

**RRB Junior Engineer (2019) - General Chemistry for CBT-1 & CBT-2** Onlineverdan,2019-04-16 This Book encompasses all topics of General Chemistry according to syllabus of CBT 1 and CBT 2 of RRB Junior Engineer 2019 Exam The whole syllabus of General Chemistry is divided in eight sections First section of basics cover topic related to classification of Chemistry matter and its composition structure of atom periodic table chemical reactions unit systems etc Second and third sections describes about bonding between oxygen and hydrogen and carbon and nitrogen respectively Fourth section describes properties of various metals properties of acids and bases importance of nanotechnology in today s scenario Fifth section discusses about wider use of chemistry in agriculture food and medical sectors Sixth section is dedicated to polymers and its various varieties available in market Seventh section is related to chemical composition of fats and proteins The last eighth section detailed out thermodynamics and gas laws given by several scientists Further each section is divided in sub sections consisting detailed theory and practice questions The level of questions are easy to tough so that students may prepare not only for this exam but also other competitive exams such as UPSC CSAT State PSCs SSC JE etc The team OnlineVerdan have shown their best efforts to bring this unique book on e publication platform **Complete Solutions and Answers for General Chemistry** John Mays,2018-01-29 All answers to exercises quizzes and tests plus worked out solutions General Chemistry uses a mastery learning method In this method students build comprehension by adding new concepts while reviewing and rehearsing key material throughout the year This method is implemented in carefully crafted exercises quizzes and the textbook narrative and it facilitates learning mastery and retention Students appreciate the smaller profile and lighter weight of our books This is possible because of aspects of our Textbook Philosophy When mastery is the goal you must cull down the material to an amount that can reasonably be mastered in one year Related subjects are integrated into the narrative The history of modern chemistry mathematics and technical communication is emphasized throughout Integration protects a course from seeming compartmentalized from other subjects That s certainly not how the real world is Real chemists use math and writing skills every day Their knowledge about how the scientific enterprise works is greatly enhanced by their knowledge of great scientists who came before Our approach to implementing a Kingdom Perspective does not entail unnaturally inserting Bible verses or devotional insets A better approach for Christians is to do science from the faith informed perspective of disciples of Christ studying God s created order This is apparent in General Chemistry from the observations about order in the universe in how nature submits to scientific study and modeling and in the fitness of creation as a habitat for humanity and animal life **Pamphlet - Dept. of the Army** United States. Department of the Army,1951 *The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success (Volume One) (First Edition* Rachel Turoscy,2018-08-09 *The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1* is a concise and accessible textbook that

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respective ions in a solution as well as the second law of Thermodynamics The text also addresses Gibbs Free Energy equilibrium constants and electrolysis calculations Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry Rachel Turoscy earned her B S M S and Ph D in chemistry from Lehigh University She is an associate professor of chemistry at Middlesex County College Dr Turoscy has authored numerous articles book chapters and textbooks addressing various topics in the world of chemistry Other Cognella titles by Rachel Turoscy The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 First Edition *Standard Nomenclature List* United States Armed Forces Institute, 1953 JEE Main 2019 Resource Book (Solved 2002 - 2018 Papers + 24 Part Tests + 10 Mock Tests) with 5 Online Tests 6th Edition Disha Experts, JEE Main 2019 Resource Book Solved 2002 2018 Papers 24 Part Tests 10 Mock Tests with 5 Online Tests 5th Edition has been divided into THREE parts Part A 24 Unit wise Tests 8 each in Physics Chemistry Mathematics Part B JEE Main AIEEE past Solved Papers 2002 2018 Papers Part C 10 Full Syllabus Mock Tests 5 in the book and 5 ONLINE empowered with Insta Results and Feedback Reports Thus all in all it is a 100% solution for both Online and Offline JEE Main exam **Master the PCAT** Peterson's, 2012-07-15 Peterson's Master the PCAT is an in depth review that offers thorough preparation for the computer based exam After learning about the structure format scoring and score reporting and the subtests and question types you can take a diagnostic test to learn about your strengths and weaknesses The next six parts of the eBook are focused on detailed subject reviews for each subtest verbal ability reading comprehension biology chemistry quantitative ability and writing Each review includes practice questions with detailed answer explanations You can take two practice tests to track your study progress The tests also offer detailed answer explanations to further improve your knowledge and understanding of the tested subjects The eBook concludes with an appendix that provides helpful information on a variety of careers in pharmacy and ten in depth career profiles Math Review Toolkit for General Chemistry Gary Long, 2006-05-30 JEE Main 2018 Resource Book (Solved 2002 - 2017 Papers + 24 Part Tests + 10 Mock Tests) with 5 Online Tests 5th Edition Disha Experts, 2017-08-30 JEE Main 2018 Resource Book Solved 2002 2017 Papers 24 Part Tests 10 Mock Tests with 5 Online Tests 5th Edition has been divided into THREE parts Part A 24 Unit wise Tests 8 each in Physics Chemistry Mathematics Part B JEE Main AIEEE past Solved Papers 2002 2017 Papers Part C 10 Full Syllabus Mock Tests 5 in the book and 5 ONLINE empowered with Insta Results and Feedback Reports Thus all in all it is a 100% solution for both Online and Offline JEE Main exam General Chemistry Darrell D. Ebbing, Larry K. Krannich, Joan I. Senyk, 1996 **Answer Key for General Chemistry** Staff of Christ the King Books, 2017-02 *New Pattern NTA JEE Main 2020 Resource Book (Solved 2002 - 2019 Papers + 24 Part Tests + 10 Mock Tests) with 5 Online Tests 7th Edition* Disha Experts, 2019-09-25 General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999 **Handbook of Academic**

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