

Exam I Study Guide

- **Physical Changes vs. Chemical Changes (Can you undo it?)**
 - Physical changes result in changes of appearance, but not composition.
 - Melting, evaporation, freezing, etc.
 - Chemical changes result in changes of composition of matter. (Chemical reaction)
 - Burning, mixing, dissolving, etc.
- **Physical Properties vs. Chemical Properties**
 - Physical properties can be described by observing the matter without a change in the composition.
 - Color, odor, density, temperature, etc.
 - Chemical properties can only be detected by altering the composition.
 - Flammability, oxidization, etc.
- **Atoms vs. Molecules**
 - Atoms are the smallest unit of a given element that carry the properties of that element.
 - Molecules are composed of two or more atoms.
- **Elements vs. Compounds**
 - Elements occur in nature and cannot be broken down to simpler particles through ordinary means.
 - Compounds are the combination of two or more elements.
- **Accuracy vs. Precision**
 - Accuracy refers to the agreement of a measurement with a known or accepted value. (Bullseye)
 - Correctness
 - Precision refers to the agreement of a measurement with other measurements in a given set of measurements. (Missing the bullseye but landing all darts in one place)
 - Repeatability or reproducibility
- **Rules Concerning Significant Figures**
 - All nonzero digits ARE significant.
 - Zeros between nonzero digits ARE significant.
 - Zeros at the beginning of a number are NOT significant; they only indicate the position of the decimal point.
 - Zeros at the end of a number to the right of a decimal point ARE significant.
 - Zeros at the end of a number with no decimal point are ASSUMED to be NOT significant, UNLESS we indicate their significance by placing a line over the last zero that is significant.

General Chemistry Standardized Final Exam Study Guide

Kathleen Armour



General Chemistry Standardized Final Exam Study Guide:

Whispering the Strategies of Language: An Emotional Quest through **General Chemistry Standardized Final Exam Study Guide**

In a digitally-driven earth where displays reign great and immediate interaction drowns out the subtleties of language, the profound strategies and mental nuances hidden within phrases frequently get unheard. However, set within the pages of **General Chemistry Standardized Final Exam Study Guide** a interesting fictional value blinking with raw thoughts, lies a fantastic journey waiting to be undertaken. Composed by a skilled wordsmith, that charming opus encourages viewers on an introspective trip, gently unraveling the veiled truths and profound impact resonating within ab muscles fabric of each and every word. Within the emotional depths with this emotional review, we will embark upon a honest exploration of the book is key themes, dissect their charming publishing design, and fail to the effective resonance it evokes deep within the recesses of readers hearts.

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