



# Loewe Xelos A42 A37 A32 Hd 100 User Guide

**K Payea**



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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you  $23 \text{ Fe} + \text{Co}^{2+} \rightarrow \text{Fe}^{2+} + \text{Co}$

oxidation  $\text{Co} + \text{Fe}^{2+} \rightarrow \text{Co}^{2+} + \text{Fe}$  reduction  $\text{Fe} + \text{Co}^{2+} \rightarrow \text{Fe}^{2+} + \text{Co}$  24 3 ag ni  $\text{Ni} + \text{Ag}^{+} \rightarrow \text{Ni}^{2+} + \text{Ag}$  oxidation reduction 25  $\text{Cu} + \text{Pb}^{2+} \rightarrow \text{Cu}^{2+} + \text{Pb}$  oxidation reduction 26  $\text{O}_2 + 2 \text{H}_2 \rightarrow 2 \text{H}_2\text{O}$

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